









## Waters Column Selection Guide for Polar Compounds

Waters HILIC and reversed-phase LC columns specifically developed to tackle polar analyte separations.

# Compound Polarity

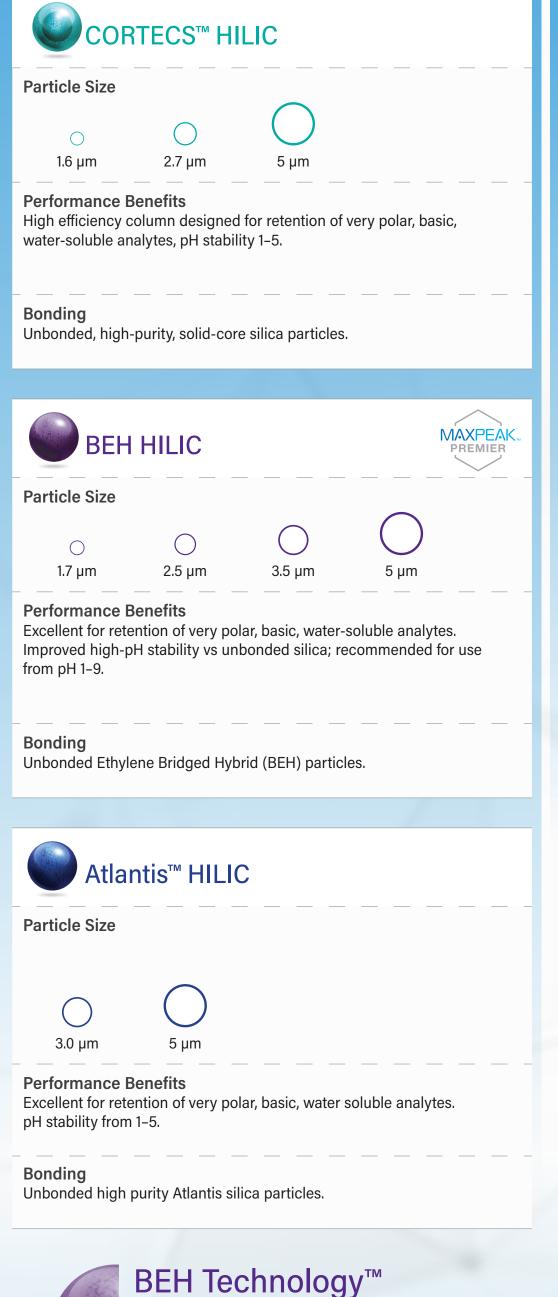
log P

More Polar Analytes HILIC Recommended

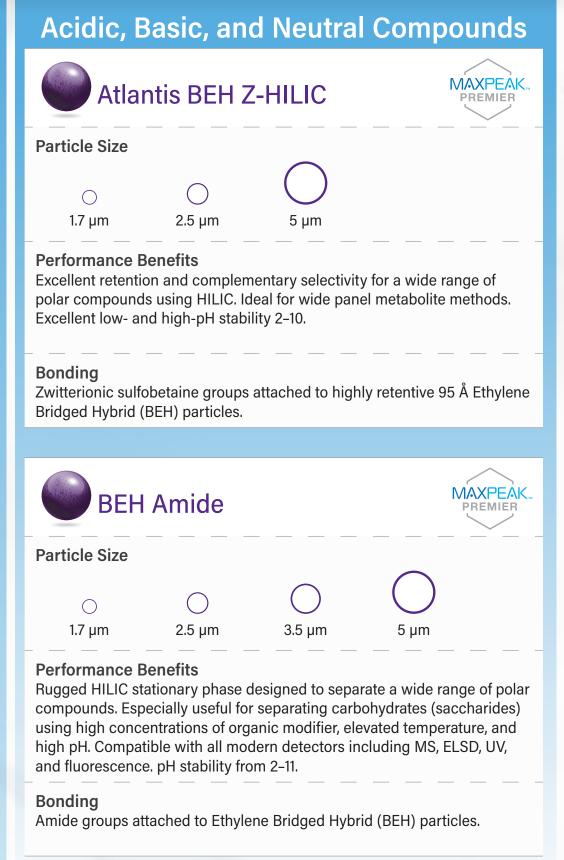
Less Polar Analytes

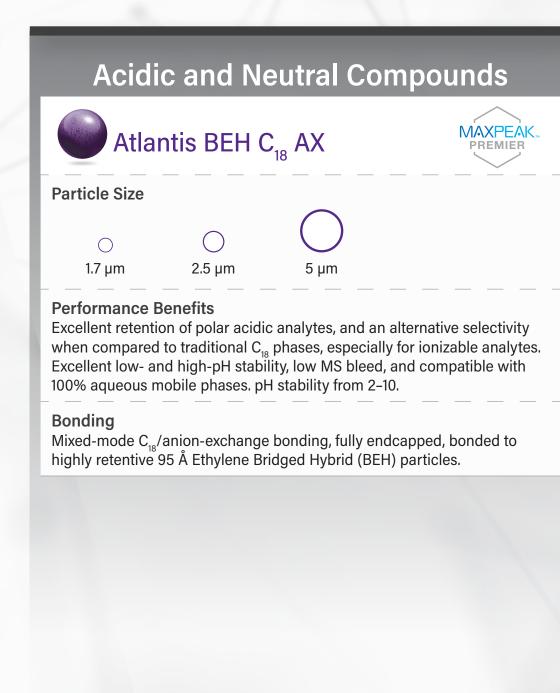
Reversed-Phase Recommended

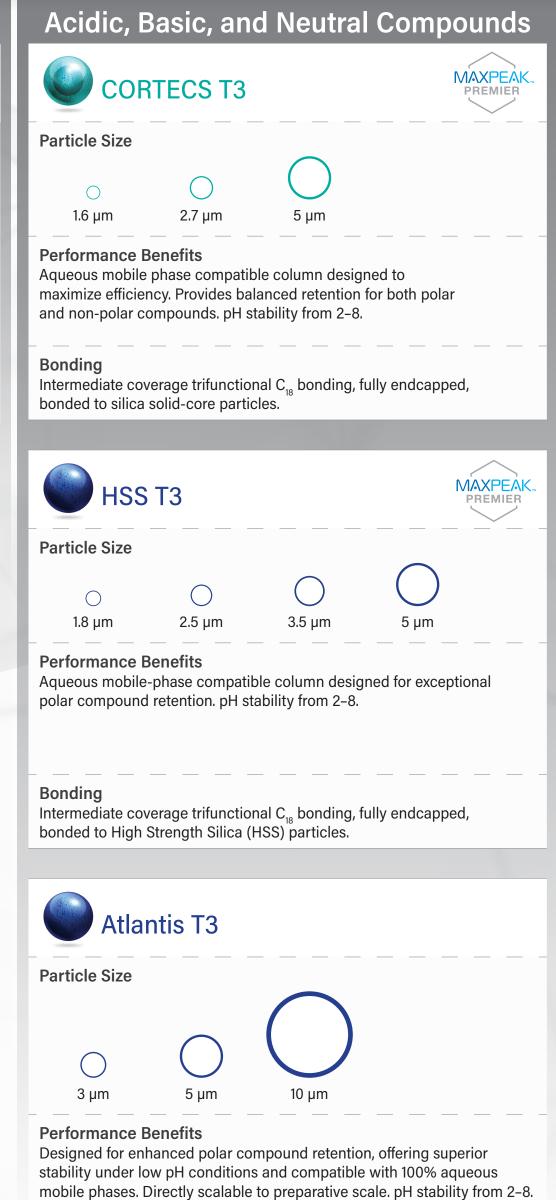
Is the polar compound acidic, basic, or neutral?



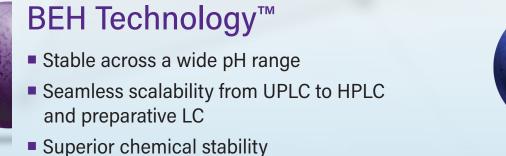
**Basic and Neutral Compounds** 







### Particle Technologies





#### **HSS Technology**

- High retentive UPLC/UHPLC/HPLC silica particle
- High strength silica for mechanical stability at elevated operating pressures
- Seamless scalability from UPLC to HPLC
- and preparative LC



#### Solid-Core Technology

- Maximum separation efficiency
- Sharper peaks for increased sensitivity Seamless scalability from UPLC to
- **UHPLC** to HPLC



#### **Atlantis Silica**

bonded to high purity Atlantis silica particles.

Intermediate coverage trifunctional C<sub>18</sub> bonding, fully endcapped,

- High retentive HPLC silica particle
- Seamless scalability from HPLC to preparative LC

### Common LC Buffers

Mobile-Phase Chemical	pK <sub>a</sub>	<b>Buffer Range</b>	Formula	Volume or Mass Required for 10 mM Mobile-Phase Concentration (per 1 L)	pH Adjustment Acid/Base	MS Compatible?	HILIC Compatible?
Acetic Acid (glacial)	4.8	_	CH₃COOH	0.571 mL		<del>-</del>	<del>√</del>
Ammonium Acetate pK <sub>a</sub> 1	4.8	3.8-5.8	CH <sub>3</sub> COONH <sub>4</sub>	0.770 g	CH₃COOH or NH₄OH	✓	✓
Ammonium Acetate pK <sub>a</sub> 2	9.2	8.2-10.2	CH <sub>3</sub> COONH <sub>4</sub>	0.770 g	CH₃COOH or NH₄OH	✓	✓
Ammonium Bicarbonate	9.2, 10.3	8.2-11.3	NH <sub>4</sub> HCO <sub>3</sub>	0.790 g	HCOOH or NH₄OH	✓	✓
Ammonium Formate pK <sub>a</sub> 1	3.8	2.8-4.8	HCOONH <sub>4</sub>	0.640 g	HCOOH or NH₄OH	✓	✓
Ammonium Formate pK <sub>a</sub> 2	9.2	8.2-10.2	HCOONH <sub>4</sub>	0.640 g	HCOOH or NH₄OH	✓	✓
Ammonium Hydroxide (28%)	9.2	_	$NH_4OH$	0.675 mL	_	✓	✓
Ammonium Phosphate, Dibasic	7.2, 9.2	6.2-10.2	$(NH_4)_2HPO_4$	1.32 g	H <sub>3</sub> PO <sub>4</sub> or NH <sub>4</sub> OH	×	✓
Formic Acid	3.8	_	НСООН	0.420 mL	_	✓	✓
N-Methylpyrrolidine	10.3	_	$C_5H_{11}N$	1.04 mL	_	✓	✓
Phosphoric Acid	2.1	_	$H_3PO_4$	0.580 mL	_	×	✓
Potassium Phosphate, Monobasic	2.1	1.1-3.1	KH <sub>2</sub> PO <sub>4</sub>	1.36 g	H <sub>3</sub> PO <sub>4</sub> or KOH	×	×
Potassium Phosphate, Dibasic	7.2	6.2-8.2	K <sub>2</sub> HPO <sub>4</sub>	1.74 g	H <sub>3</sub> PO <sub>4</sub> or KOH	×	×
Potassium Phosphate, Tribasic	12.7	11.7–13.7	$K_3PO_4$	2.12 g	H <sub>3</sub> PO <sub>4</sub> or KOH	×	×
Pyrrolidine	11.3	_	$C_4H_9N$	0.833 mL	_	✓	✓
Sodium Borate	9.1, 12.7, 13.8	8.2–14	$Na_2B_4O_7$	2.01 g	H <sub>3</sub> BO <sub>4</sub> or NaOH	×	×
Sodium Citrate, Tribasic	3.1, 4.8, 6.4	2.1-7.4	HOC(COONa)(CH2COONa)2	2.58 g	Citric Acid or NaOH	×	×
Triethylamine (TEA)	11.01	_	(CH3CH2)3N	1.39 mL	_	✓	✓
Triethylammonium Acetate (TEAA) pK <sub>a</sub> 1	4.76	3.8-5.8	(CH <sub>3</sub> CH <sub>2</sub> ) <sub>3</sub> NH:CH <sub>3</sub> COO	0.695 mL TEA/0.571 mL Acetic Acid	TEA or CH₃COOH	✓	✓
Triethylammonium Acetate (TEAA) pK <sub>a</sub> 2	11.01	10.0-12.0	(CH <sub>3</sub> CH <sub>2</sub> ) <sub>3</sub> NH:CH <sub>3</sub> COO	1.39 mL TEA/0.285 mL Acetic Acid	TEA or HCOOH	✓	✓
Triethylammonium Formate (TEAF) pK <sub>a</sub> 1	3.75	2.8-4.8	(CH <sub>3</sub> CH <sub>2</sub> ) <sub>3</sub> NH:HCOO	0.695 mL TEA/0.420 mL Formic Acid	TEA or HCOOH	✓	✓
Triethylammonium Formate (TEAF) pK <sub>a</sub> 2	11.01	10.0-12.0	(CH <sub>3</sub> CH <sub>2</sub> ) <sub>3</sub> NH:HCOO	1.39 mL TEA/0.210 mL Formic Acid	TEA or HCOOH	✓	✓
Trifluoroacetic Acid (TFA)	0.3	_	CF₃COOH	0.743 mL	_	✓	<b>√</b>



Select column configurations that show the MaxPeak Premier symbol are available in the MaxPeak Premier Column format. The MaxPeak Premier Columns utilize MaxPeak High Performance Surfaces (HPS) Technology which increases reproducibility, improves peak shape, and enables more accurate recovery by minimizing unwanted analyte/surface interactions.

