

#### Application Note

# Reliable LC/UV Analysis of Nitrite and Nitrate Using the Arc™ Premier System With an Atlantis™ Premier BEH™ C<sub>18</sub> AX Column

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#### Abstract

Nitrite and nitrate are considered as potential precursors in the formation of nitrosamine impurities, carcinogenic compounds that can cause cancer. Herein, we present a reverse phase method developed for the analysis of nitrite and nitrate using an Arc Premier System with UV detection and an Atlantis Premier BEH  $C_{18}$  AX Column. The method is based on direct injection, eliminating the need for a derivatization step prior to analysis. The limits of quantitation (LOQ) for nitrite and nitrate were found to be 0.025  $\mu$ g/mL and 0.075  $\mu$ g/mL, respectively. The method demonstrated excellent repeatability, precision, and linearity and was found to be suitable for the determination of nitrite and nitrate in the polyvinylpyrrolidone (PVP) excipient with accurate quantitative performance with 91–100% recovery.

#### **Benefits**

- Successful retention of the nitrite and nitrate ions under reverse phase conditions using an Atlantis Premier
   BEH C<sub>18</sub> AX Column
- Fast and easy screening method by direct injection, eliminating the need for a complex sample derivatization step prior to analysis

## Introduction

Nitrite and nitrate are naturally occurring nitrogen compounds in soil, water, air, and plants.<sup>1</sup> They are also used in food preservation, some pharmaceutical drugs, and in the production of explosives.<sup>1</sup> Recent publications have indicated that nitrite and nitrate are considered precursors or nitrosating agents in the formation of N-nitrosamine compounds, potential human carcinogens.<sup>2-6</sup> Nitrosamines are generally formed by the reaction of secondary or tertiary amines in the pharmaceutical active pharmaceutical ingredients (APIs) with the residual nitrite during the drug product manufacturing process. Nitrate is also recognized as nitrosating agent as it can be reduced to nitrite under certain conditions.<sup>4</sup> Formulation ingredients, including excipients, can contain nitrosating precursors, which may contribute to nitrosamine contamination in drug products.<sup>4</sup>

The most widely utilized methods for nitrate and nitrite determination involve the use of ion chromatography (IC) with conductivity detection, liquid chromatography (LC) with UV detection, and the Griess test.<sup>4–6</sup> While effective, these methods allow the determination of a single analyte, nitrite, or nitrate, and often require different sample preparation procedures. The Griess test is specific for nitrite and cannot be used to measure nitrate content.<sup>4</sup> Considering the possible conversion of nitrate to nitrite upon reduction, there is a desire to measure both compounds.

Herein, a single liquid chromatography (LC) method with UV detection is described for the analysis of both nitrite and nitrate, based on direct injection. This reverse phase method offers easy and fast screening for nitrite and nitrate in polyvinylpyrrolidone excipient.

# Experimental

High purity LC/MS grade acetonitrile, ammonium formate, and formic acid were purchased from Fisher Chemicals. Sodium nitrate (NaNO<sub>3</sub>) and sodium nitrite (NaNO<sub>2</sub>) analytical grade quality were purchased from Sigma. Polyvinylpyrrolidone (PVP) K 15 was obtained from Sigma.

#### Standard Solutions

Nitrite and nitrate stock solutions were prepared by dissolving sodium nitrite and sodium nitrate salts in water at 5.0 mg/mL. Stock solutions were diluted with water to make a mixture standard solution containing nitrite and

nitrate at 0.1 mg/mL concentration. Calibration standards were prepared by diluting the mixture standard solution with water.

# Polyvinylpyrrolidone (PVP) Sample Solutions

Test sample solutions were prepared in water at 1 mg/mL and dissolved by sonication for 20 minutes. After extraction, sample solutions were centrifuged for ten minutes at 3000 rpm and filtered through a PVDF 0.2 µm syringe filter (p/n: WAT200806 <a href="https://www.waters.com/nextgen/us/en/shop/sample-preparation-filtration/wat200806-acrodisc-syringe-filter-pvdf-13-mm-02--m-aqueous-100-pk.html">https://www.waters.com/nextgen/us/en/shop/sample-preparation--filtration/wat200806-acrodisc-syringe-filter-pvdf-13-mm-02--m-aqueous-100-pk.html</a> ) prior to analysis.

## **Method Conditions**

LC system:	Arc Premier System, column manager with active pre-heating, PDA Detector
Vials:	LC/MS Maximum Recovery 2 mL volume, p/n: 600000670CV
Detection:	UV at 210 nm
Column:	Atlantis Premier BEH $C_{18}$ AX, 2.1 x 100 mm, 2.5 $\mu$ m (p/n: 186009392)
Column temperature:	30 °C
Sample temperature:	15 °C
Injection volume:	40.0 μL
Flow rate:	0.5 mL/min
Mobile phase A:	5 mM Ammonium formate in water with 0.1% formic acid in water

Mobile phase B: Acetonitrile with 0.1% formic acid

Separation: Isocratic separation with 70:30 mobile phase

A/mobile phase B

# Data Management

Chromatography software: Empower™ 3 Feature Release 5 Service Release 5

(FR5 SR5)

Data acquisition and analysis performed using Empower Software. Summary reports generated using the report templates are available in the Empower project.

## Results and Discussion

Various column chemistries, organic solvents, mobile phase additives, and pH were explored during the method development to ensure chromatographic retention and separation of nitrite and nitrate ions. The Atlantis Premier BEH  $C_{18}$  AX Column successfully retained and separated nitrite and nitrate under reverse-phase conditions (Figure 1A). The replicate injections of the 0.1  $\mu$ g/mL standard demonstrated excellent repeatability of the Arc Premier System for nitrite and nitrate analysis with relative standard deviation (RSD) for peak areas of 0.94 and 3.06%, respectively (Figure 1B).

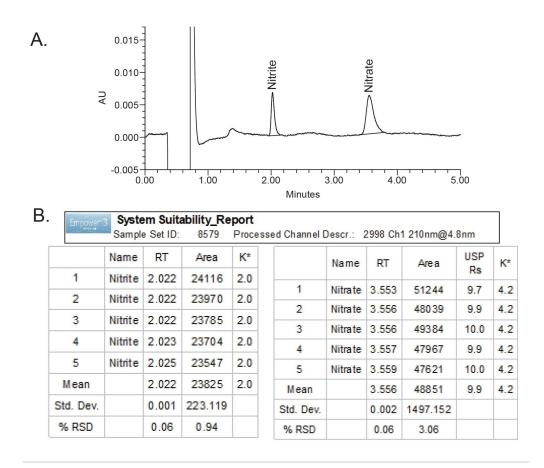


Figure 1. Nitrite and nitrate standard at 0.1  $\mu$ g/mL run on an Arc Premier System with an Atlantis Premier BEH  $C_{18}$  AX Column (A) and system suitability results for replicate injections (n=5).

# Limits of Detection and Quantitation (LOD and LOQ)

The minimum concentration levels at which nitrite and nitrate can be reliably detected and quantified using the UV method were determined using the signal-to-noise (S/N) criteria of 3:1 and 10:1, respectively. Calculation of S/N ratio was performed using Empower software by comparing the measured signal for nitrite and nitrate in the standard solution to the blank solution. Data from ten replicate injections of standard solutions was used to establish and assess performance of the method at the LOD and LOQ levels.

Representative chromatograms showing the analysis of nitrite and nitrate LOD and LOQ solutions are shown in Figure 2. The method performance at the LOD and LOQ levels is shown in Figure 3. The LOD and LOQ for nitrite

was found to be 0.01 and 0.025  $\mu$ g/mL, respectively. The LOD and LOQ for nitrate was at 0.05 and 0.075  $\mu$ g/mL, respectively. The data from ten replicate injections at the LOQ levels demonstrated excellent repeatability of the method, with RSD of peak areas at 4.30% for nitrite and 5.52% for nitrate (Figure 3).

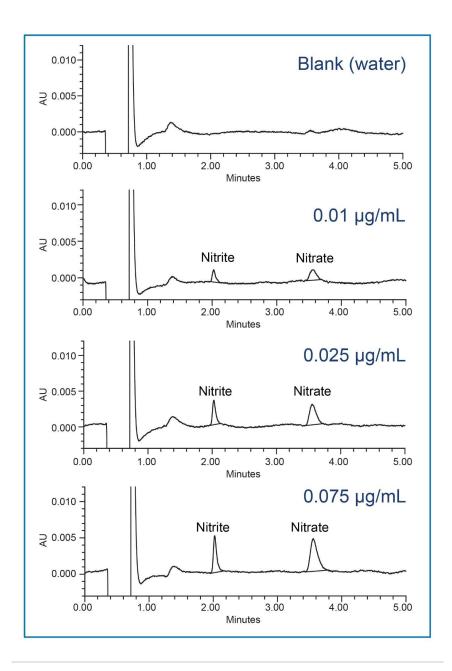


Figure 2. Representative chromatograms of nitrite and nitrate LOD and LOQ solutions, UV at 210 nm.

Nitrite (NO <sub>2</sub> )							Nitrate (NO₃)								
LO	D: 0.0	01 μg/m	nL	LO	Q: 0.	025 µg	/mL	LOD: 0.05 μg/mL LOQ: 0.075 μg/				mL			
	RT	Area	USP s/n		RT	Area	USP s/n		RT	Area	USP s/n		RT	Area	USP s/n
1	2.020	1970	4	1	2.020	5658	10	1	3.552	23513	3	1	3.550	34429	10
2	2.022	2155	4	2	2.021	5534	10	2	3.553	21191	3	2	3.552	36517	10
3	2.022	2375	4	3	2.022	5137	10	3	3.555	22444	3	3	3.553	33639	10
4	2.023	2271	4	4	2.023	5481	10	4	3.555	20874	3	4	3.554	33332	10
5	2.024	2328	4	5	2.023	5668	10	5	3.556	27695	4	5	3.555	34830	10
6	2.025	2186	4	6	2.024	5541	10	6	3.557	22186	3	6	3.556	35810	10
7	2.027	2416	4	7	2.025	5217	10	7	3.561	26177	3	7	3.558	38732	10
8	2.032	2347	3	8	2.025	5236	10	8	3.561	22398	3	8	3.560	38149	10
9	2.033	2054	3	9	2.026	5797	8	9	3.562	26379	3	9	3.560	33453	10
10	2.038	2562	3	10	2.027	5187	8	10	3.566	19151	3	10	3.560	37001	10
Mean	2.027	2266	4	Mean	2.024	5446	10	Mean	3.558	23201	3	Mean	3.556	35589	10
Std. Dev.	0.006	177.578		Std. Dev.	0.002	234.425		Std. Dev.	0.004	2735.710		Std. Dev.	0.003	1964.001	
% RSD	0.28	7.84		% RSD	0.10	4.30		% RSD	0.12	11.79		% RSD	0.10	5.52	

Figure 3. Method performance at the LOD and LOQ levels. UV at 210 nm.

# Linearity and Concentration Range

The calibration plot for nitrite and nitrate was constructed in the range of 0.025 to 5  $\mu$ g/mL and 0.05 to 5  $\mu$ g/mL, respectively. The calibration curve of the peak areas versus concentration produced a linear relationship with a correlation coefficient (R<sup>2</sup>)  $\geq$ 0.999 (Figure 4).

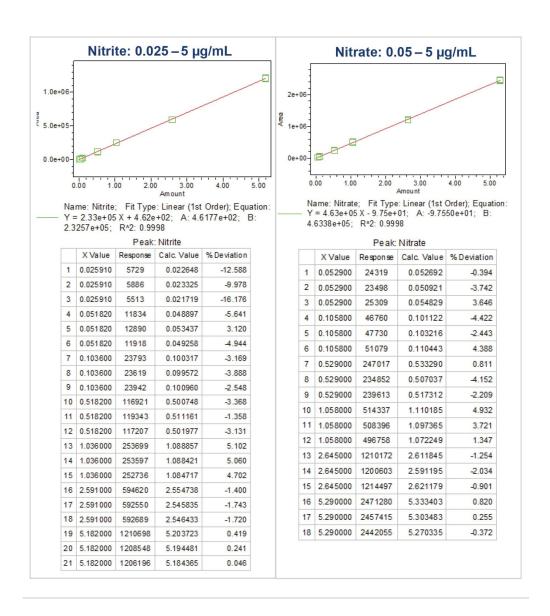


Figure 4. Calibration curve for nitrite and nitrate. UV at 210 nm.

# Polyvinylpyrrolidone Analysis

Polyvinylpyrrolidone, also known as povidone or PVP, is a synthetic polymer used in the pharmaceutical industry as a binder in tablet and capsules manufacturing, a film former for ophthalmic solutions, to aid in flavoring liquids and chewable tablets, and as an adhesive for transdermal systems.<sup>7</sup>

The PVP test samples, prepared at 1 mg/mL in water, were analyzed using the developed method.

Representative chromatograms of the PVP sample and the PVP sample spiked with nitrite and nitrate are shown in Figure 5. The method effectively separated the highly concentrated PVP peak from the nitrite and nitrate. In the PVP test sample, no nitrite was detected, but residual levels of nitrate were found, at a concentration of 0.04  $\mu$ g/mL.

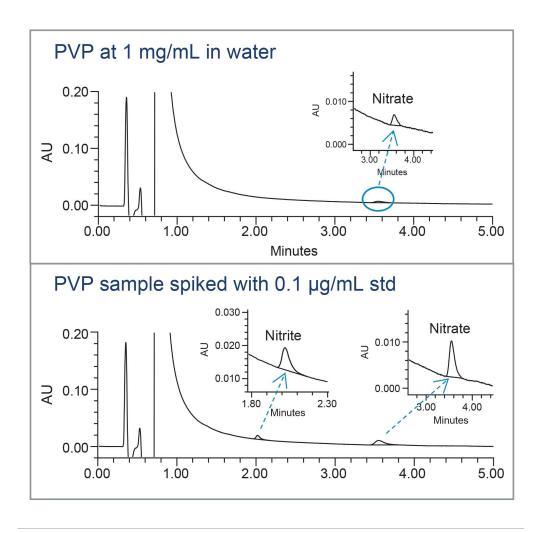


Figure 5. Analysis of polyvinylpyrrolidone (PVP) test sample solutions. UV at 210 nm.

# Method Accuracy

The accuracy of the method was assessed by determining recovery of the nitrite and nitrate spiked into PVP samples at three levels including 0.05, 0.1, and 0.5  $\mu$ g/mL. For nitrate recovery, calculations were corrected for

the amount already present in the sample. Recoveries for nitrite and nitrate (n=6 per each level) ranged from 93 to 100% and 91 to 100% (Figure 6), respectively, within the acceptance criteria range of 90–110%.

#### % Recovery at 0.05 µg/mL

#### % Recovery at 0.1 µg/mL

#### % Recovery at 0.5 µg/mL

	SampleName	Nitrite
1	PVP 1 mg +0.05ug	96.8
2	PVP 1 mg +0.05ug	95.6
3	PVP 1 mg +0.05ug	98.2
4	PVP 1 mg +0.05ug	95.3
5	PVP 1 mg +0.05ug	93.2
6	PVP 1 mg +0.05ug	96.4
Mean		96
Std. Dev.		1.685
% RSD		1.76

	SampleName	Nitrite	Nitrate
1	PVP 1 mg +0.1ug	93.8	95.6
2	PVP 1 mg +0.1ug	95.6	90.6
3	PVP 1 mg +0.1ug	94.7	91.9
4	PVP 1 mg +0.1ug	98.3	96.2
5	PVP 1 mg +0.1ug	100.0	91.9
6	PVP 1 mg +0.1ug	99.7	91.9
Mean		97	93
Std. Dev.		2.666	2.326
% RSD		2.75	2.50

	SampleName	Nitrite	Nitrate
1	PVP 1 mg +0.5ug	97.6	99.8
2	PVP 1 mg +0.5ug	98.2	95.2
3	PVP 1 mg +0.5ug	100.1	95.4
4	PVP 1 mg +0.5ug	98.7	96.0
5	PVP 1 mg +0.5ug	98.4	96.4
6	PVP 1 mg +0.5ug	100.3	98.1
Mean		99	97
Std. Dev.		1.076	1.793
% RSD		1.09	1.85

Figure 6. Method accuracy showing percent (%) recovery for nitrite and nitrate spiked to PVP test samples.

# Conclusion

A single LC/UV with direct injection method was developed for the analysis of both nitrite and nitrate, utilizing the Arc Premier System and an Atlantis Premier BEH  $C_{18}$  AX Column. The method demonstrated excellent performance at the LOQ levels of  $0.025~\mu g/mL$  and  $0.075~\mu g/mL$  for nitrite and nitrate respectively. The method has been demonstrated as suitable for the analysis of polyvinylpyrrolidone excipient, with recoveries of nitrite and nitrate between 91 to 100%. The LC-UV method offered a quick and easy solution for screening nitrite and nitrate in the PVP K15 excipient or raw materials.

## References

1. https://wwwn.cdc.gov/TSP/ToxFAQs/ToxFAQsDetails.aspx?faqid=1186&toxid=258 <

https://wwwn.cdc.gov/TSP/ToxFAQs/ToxFAQsDetails.aspx?faqid=1186&toxid=258>.

2. European Medicine Agency, Procedure under Article 5(3) of Regulation EC (No) 726/2004. Nitrosamines impurities in human medicinal products EMA/369136/2020.

3. FDA Guidance document, Control of nitrosamine impurities in human drugs, guidance for industry.

4. Boetzel R, Schlingemann J, Hickert S, Korn C, Kocks G, Luck B, Blom G, Harrison M, Francois M, Allain L, Wu Yongmei, Bousraf Y. A nitrite excipient database: A useful tool to support N-nitrosamine risk assessments for drug products. *Journal of Pharmaceutical Sciences*, (2022) 1–10.

Holzgrabe U. Nitrosated Active Pharmaceutical Ingredients – Lessons Learned. *Journal of Pharmaceutical Sciences*, 000 (2023) 1–6.

6. Jires J, Dousa M. Nitrites as precursors of N-nitrosation in pharmaceutical samples – A trace level analysis. *Journal of Pharmaceutical and Biomedical Analysis*, 213 (2022) 114677.

7. https://www.drugs.com/inactive/polyvinylpyrrolidone-271.html 
https://www.drugs.com/inactive/polyvinylpyrrolidone-271.html>.

#### Featured Products

Arc Premier System <a href="https://www.waters.com/nextgen/us/en/products/chromatography/chromatography-systems/arc-premier-system.html">https://www.waters.com/nextgen/us/en/products/chromatography/chromatography-systems/arc-premier-system.html</a>

Empower Chromatography Data System <a href="https://www.waters.com/10190669">https://www.waters.com/10190669</a>>

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