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アプリケーションノート

# Routine Quantitation of PAHs in Sediments Using Electron Ionization GC-MS/MS

Kari Lynn Organtini, Stephanie Crombie, Serge Moore, Kenneth Rosnack

日本ウォーターズ株式会社



This is an Application Brief and does not contain a detailed Experimental section.

**Abstract** 

This application brief evaluates the Xevo TQ-GC as a fit for purpose electron ionization (EI) GC-MS/MS for the routine analysis of polycyclic aromatic hydrocarbons (PAHs) extracted from sediment samples.

#### **Benefits**

Xevo TQ-GC is a fit for purpose GC-MS/MS solution for routine analysis of PAHs in sediment samples.

#### Introduction

Polycyclic aromatic hydrocarbons (PAHs) are common persistent organic pollutants (POPs) monitored by many agencies globally. PAHs can be generated naturally and anthropogenically through combustion of carbon-based fuels such as wood, coal, oil, and diesel, making them present in food, water, and soil samples worldwide. They are made up of fused aromatic rings and are classified as being carcinogenic or potentially carcinogenic. Although there are hundreds of native and substituted known PAHs, only a handful are typically focused on in regulatory methods. For example, Canadian Soil Quality Guidelines<sup>1</sup> recommend an analysis of 41 PAHs, Europe<sup>2</sup> follows the 15+1 list, and the United States<sup>3</sup> tends to follow a list of 16 priority EPA PAHs. Xevo TQ-GC is a fit for purpose GC-MS/MS solution for routine analysis of PAHs in sediment samples.

The goal of this evaluation was to expand the list of PAHs in one GC-MS/MS analysis to include the majority of the PAHs required in Canada, Europe, and the United States. Various sediment sample extracts previously extracted on site at Environment and Climate Change Canada Quebec Laboratory for Environmental Testing (QLET)<sup>4</sup> were used for this evaluation. A suite of approximately 50 PAHs (C9 through C24 PAHs as well as chlorinated and methylated isomers) were evaluated on the Xevo TQ-GC for routine performance criteria such as sensitivity, calibration range, and accuracy based on analysis of the extracted sediment samples.

#### **Results and Discussion**

Chromatographic performance was crucial for this analysis as the suite of PAHs included in the method contains three groups of critical isomer separations that need to be resolved; phenanthrene/anthracene, 2/3/4/5/6-methyl chrysene, and benzo(b)/(j)/(k)fluoranthene. Utilizing the Rxi-35Sil MS column, all the critical isomers were separated (Figure 1) sufficiently to achieve reliable quantitation.

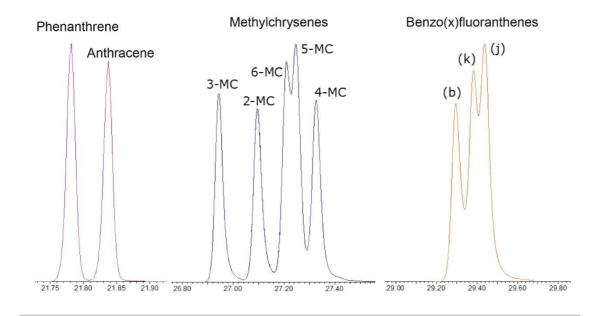


Figure 1. Three critical isomer pairs resolved using the Rxi-35Sil MS column. Full method details can be accessed and downloaded from the Xevo TQ-GC Quanpedia Database (marketplace.waters.com).

The Xevo TQ-GC demonstrated good calibration over the calibration range used (0.2–40  $\mu g/kg$ ) for sample analysis (Figure 2), with all R² values >0.998. As demonstrated in Figure 2, some PAHs produced a linear calibration curve while others had a slightly quadratic nature. The best fit was used for each compound in order to perform accurate quantitation. The detection limits for most compounds analyzed were determined to be below the lowest calibration point of 0.2  $\mu g/kg$  (ppb). Table 1 highlights the detection limits for each compound in the analysis method.

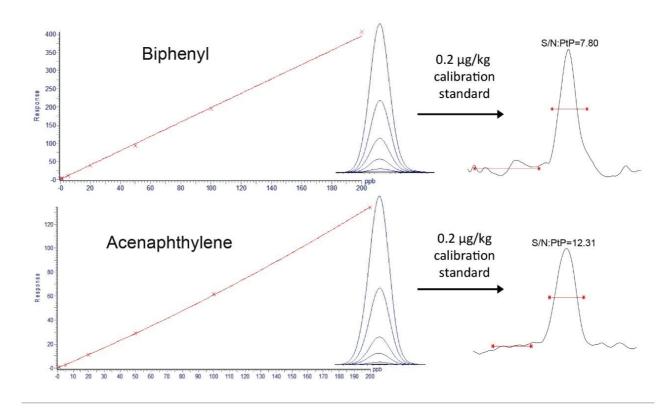


Figure 2. Calibration curves and peak overlays of each calibration point for biphenyl (top) and acenaphthylene (bottom).

| Compound   | Limit of Detection<br>(μg/kg) | Canadian Soil Quality<br>Guidelines0 | EU 15+1 List* | EPA 16 List |
|--|-------------------------------|--------------------------------------|---------------|-------------|
| Indene   | 0.04                          |                                      |               |             |
| Naphthalene  | 0.12                          | Х                                    |               | х           |
| 1-Methylnaphthalene  | 0.07                          |                                      |               |             |
| 2-Methylnaphthalene  | 0.07                          |                                      |               |             |
| 1-Chloronaphthanlene   | 80.0                          |                                      |               |             |
| 2-Chloronaphthanlene   | 0.08                          |                                      |               |             |
| Biphenyl   | 0.18                          |                                      |               |             |
| 1,3 Dimethylnaphthalene  | 0.06                          |                                      |               |             |
| Acenaphthylene   | 0.04                          |                                      |               | ×           |
| Acenaphthene   | 0.1                           |                                      |               | х           |
| 2,3,5 Trimethylnaphthalene   | 0.1                           |                                      |               |             |
| Fluorene   | 0.12                          | X                                    |               | х           |
| Phenanthrene   | 0.1                           | X                                    |               | ×           |
| Anthracene   | 0.16                          | x                                    |               | ×           |
| Carbazole  | 0.18                          | x                                    |               | ,10)        |
| 1-Methylphenanthrene   | 0.12                          | ×                                    |               |             |
| Fluoranthene   | 0.08                          | x                                    |               | х           |
| Pyrene   | 0.08                          | x                                    |               | x           |
| Retene   | 0.16                          | 53538                                |               | (7)         |
| 2 Methylfluoranthene   | 0.1                           | x                                    |               |             |
| Benzo(c)phenanthrene   | 0.1                           | x                                    |               |             |
| Benzo(c)acridine   | 0.16                          | x                                    | -             |             |
| Benzo(a)anthracene   | 0.1                           | x                                    | х             | х           |
| Chrysene   | 0.04                          | ×                                    | x             | х           |
| 2-Methylchrysene   | 0.14                          | x                                    |               | ^           |
| 3-Methylchrysene   | 0.14                          | X                                    |               |             |
| 4-Methylchrysene   | 0.14                          | x                                    |               |             |
| 5-Methylchrysene   | 0.14                          | x                                    | x             |             |
| 6-Methylchrysene   | 0.14                          | x                                    | ^             |             |
| Benzo(b)fluoranthene   | 0.14                          | 8000                                 |               | - L         |
| Benzo(k)fluoranthene   | 0.14                          | X                                    | x             | X           |
|  | 0.14                          | ×                                    | х             | Х           |
| Benzo(j)fluoranthene 7,12 dimethylbenzo(a)anthracene   | 200.000                       | X                                    | Х             |             |
| ACTION OF THE SECRETARY | 0.12                          | No.                                  |               | <u> </u>    |
| Benzo(a)pyrene   | 0.16                          | х                                    | х             | х           |
| Benzo(e)pyrene   | 0.16                          | X                                    |               |             |
| Perylene   | 0.18                          | X                                    |               |             |
| 3-Methylcholanthrene   | 0.18                          |                                      |               |             |
| Dibenzo(ah)acridine  | 80.0                          | X                                    |               |             |
| Dibenzo(aj)anthracene  | 0.06                          | X                                    |               |             |
| Dibenzo(ac)anthracene  | 0.06                          | X                                    |               |             |
| Dibenzo(ah)anthracene  | 0.06                          | X                                    | х             | Х           |
| Indeno(1,2,3-cd)pyrene   | 0.04                          |                                      | Х             | Х           |
| 7H Dibenzo(cg)carbazole  | 0.54                          | X                                    |               |             |
| Benzo(ghi)perylene   | 0.04                          | X                                    | х             | х           |
| Anthanthrene   | 0.14                          | х                                    |               |             |
| Dibenzo(ae)fluoranthene  | 0.06                          | X                                    |               |             |
| Coronene   | 0.12                          |                                      |               |             |

- 2. Joint Research Center (JRC) Technical Notes, Polycyclic Aromatic Hydrocarbons (PAHs) Factsheet, 4th edition, 2011.
- 3. EPA Method 8310, Revision 0, September 1986, Third Edition of the Test Methods for Evaluating Solid Waste, Physical/Chemical Methods, EPA publication SW 846.
- 4. Method: Q0510S0. Analyse des hydrocarbures aromatiques polycycliques dans des solides-technique pas GC-MS. Environment and Climate Change Canada Quebec Laboratory for Environmental Testing.

### **Featured Products**

Xevo TQ-GC Mass Spectrometry System <a href="https://www.waters.com/134977323">https://www.waters.com/134977323</a>

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