

应用纪要

Troubleshooting Common System Problems Using Waters Neutrals Quality Control Reference Material

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Abstract

The Waters Neutrals QC Reference Material (QCRM) is a mixture of three neutral compounds that can be separated with common mobile phases with sufficient organic composition, and is compatible with most column chemistries, making it an ideal standard for troubleshooting HPLC and UPLC system problems. In this application note, six common chromatographic problems are examined to demonstrate the utility of the Neutrals QCRM in rapidly diagnosing problems on an HPLC or UPLC system.

Benefits

- HPLC/UPLC system issues can be detected early, minimizing collection of poor quality data
- Using a quality standard, troubleshooting can be performed faster, reducing overall system downtime
- A recently repaired instrument can be quickly confirmed to be back in normal operation

Introduction

Liquid chromatography is a powerful analytical method of analysis, however, when an HPLC or UPLC system begins to malfunction, it can mean a considerable amount of time and resources to fix. Some system problems, such as a leak in the pump, can be noticed by an experienced chromatographer, while other problems, such as improperly connected column outlet tubing, can be a subtle problem and difficult to troubleshoot. By using a qualified system suitability standard, a chromatographer can more easily detect problems within their system, potentially reducing system downtime.

Waters Neutrals Quality Control Reference Material (QCRM) is a mix of three neutral compounds: acetone, naphthalene, and acenaphthene. These compounds are manufactured with batch-to-batch reproducibility in a controlled setting ensuring consistent results over time. This standard is an ideal solution for system troubleshooting and maintenance as the separation of these compounds can be achieved under common mobile phase conditions with sufficient organic content. In this application, six common chromatographic problems are examined to demonstrate the utility of the Neutrals QCRM in rapidly diagnosing problems on an HPLC or UPLC system. After repairs were made, the Neutrals QCRM was used to quickly confirm that the system was back to operating optimally. By using the Neutrals QCRM to check system functionality, data quality can be assured and the system can be used with confidence.

Experimental

ACQUITY UPLC H-Class Conditions

Mobile phase: 50:50 Acetonitrile:water

Separation mode: Isocratic

Detection (PDA): UV 254 nm

Column: ACQUITY UPLC BEH C₁₈, 2.1 x

50 mm, $1.7 \mu m$ (p/n

186002350)

Column temp.: 30 °C

Needle wash: 50:50 ACN:water

Sample purge: 50:50 ACN:water

Seal wash: 50:50 MeOH:water

Flow rate: 0.6 mL/min

Injection volume: 1 μL

Data management: Empower 3 CDS

Sample preparation

A vial of ASR Neutral QC Reference Material (p/n 186006360) was opened and transferred into an LCGC Certified Clear Qsert Vial (p/n 186001126C) for injection

Results and Discussion

When problems arise on an HPLC or UPLC system, system troubleshooting can be a time consuming and costly process. By using a QC Reference Material (QCRM), a chromatographer can more easily identify problems with a system and make repairs, potentially reducing system downtime. Waters Neutrals QCRM is a mixture of acetone, naphthalene, and acenaphthene that is manufactured in a tightly controlled setting with batch-to-batch reproducibility, to ensure reliable results. The standard can be used to both benchmark system performance and troubleshoot system problems. Furthermore, the standard can be used after repairs to ensure the system is operating optimally.

It is beneficial to have a system's performance benchmarked, in order to monitor system performance and ensure quality data generation. Subsequently, if any problems arise within the system, the operator can compare the performance after any repairs, to performance before the problem was present. In this application, six common chromatographic issues are examined, using the Neutrals QCRM to rapidly diagnose the problem, shown in Table 1. During this experiment retention time, USP tailing factor, and USP plate count were monitored. Although many other parameters may be

monitored, these parameters were chosen since many methods, especially compendial methods, have requirements for these parameters. Furthermore, all of these parameters can be indicators of a malfunctioning system.

Problem#	Problem description		
1	Poor column performance		
2	Leak in pump		
3	Poor check valve performance		
4	Improper column fitting connections		
5	Air bubble in system		
6	Error in mobile phase preparation		

Table 1. Examples of some common LC system issues intentionally induced that can be diagnosed with Waters Neutrals QCRM.

Forty-five injections of the Neutrals QCRM were run on a recently calibrated ACQUITY UPLC H-Class with ACQUITY UPLC PDA detector for five days, as shown in Table 2, prior to any system malfunction or user error. As part of a system benchmarking process, the operator should create appropriate specifications according to laboratory protocols that the Neutrals QCRM must pass in order for the system to be considered working optimally.² By creating these specifications, the system performance can be monitored, potentially catching problems before they arise.

		Acetone	Naphthalene	Acenaphthene
	Retention Time Average (min)	0.323	1.633	2.893
Benchmarking Data	Retention Time %RSD	0.690	0.440	0.440
(n=45)	USP Tailing Factor	1.180	1.130	1.080
-	USP Plate Count	3144	11009	10436

Table 2. Data gathered during system benchmarking, demonstrating a well operating system with a low retention time %RSD (n=45).

To showcase the troubleshooting capability of the Neutrals QCRM, the first issue demonstrated is the effect of a failing column on the separation. Over time, with repeated injections, all LC columns will lose their efficiency and ability to separate components of a mixture. In Figure 1, the separation of the Neutrals QCRM on an ACQUITY UPLC BEH C_{18} , 2.1×50 mm, $1.7 \mu m$ Column that had been excessively used is shown and compared to the separation obtained on a column with acceptable performance.

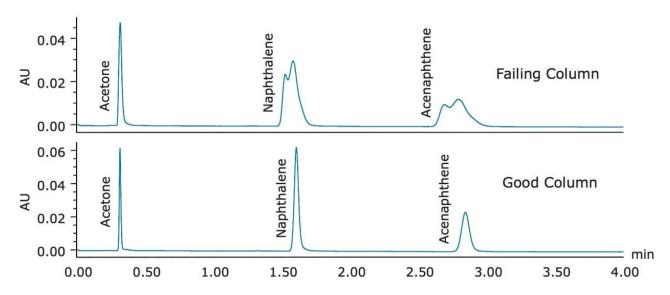


Figure 1. Separation of the Neutrals QCRM on failing column and a good column.

As Figure 1 shows, the failing column is causing peak splitting of both the naphthalene and acenaphthene peaks. Monitoring USP plate count for the acenaphthene peak, the value dropped to approximately 1000 with the failing column, shown in Table 3. After the failing column was replaced with a new column, nine injections of the Neutrals QCRM were run. The data from these nine injections, shown in Table 3, is comparable to the benchmarked data, indicating that the system is operating optimally prior to column failure. The low %RSD of the combined retention times after column replacement, as well as the consistent return to comparable plate counts and tailing factors, demonstrate that the system is back to normal performance.

	Acetone	Naphthalene	Acenaphthene
Retention Time (min)	0.319	1.574	2.770
USP Tailing Factor	1.290	1.040	1.010
USP Plate Count	1043	1458	1433
Retention Time Average (min)	0.323	1.631	2.893
Retention Time %RSD	0.670	0.460	0.440
USP Tailing Factor	1.170	1.130	1.090
USP Plate Count	3152	11001	10531
	USP Tailing Factor USP Plate Count Retention Time Average (min) Retention Time %RSD USP Tailing Factor	Retention Time (min) 0.319 USP Tailing Factor 1.290 USP Plate Count 1043 Retention Time Average (min) 0.323 Retention Time %RSD 0.670 USP Tailing Factor 1.170	Retention Time (min) 0.319 1.574 USP Tailing Factor 1.290 1.040 USP Plate Count 1043 1458 Retention Time Average (min) 0.323 1.631 Retention Time %RSD 0.670 0.460 USP Tailing Factor 1.170 1.130

Table 3. Data collected during the use of a failing column and after replacement of the column (n=9).

The second system issue demonstrated is malfunctioning pump caused by a leak. Once the leak was induced, the Neutrals QCRM was analyzed, shown in Figure 2.

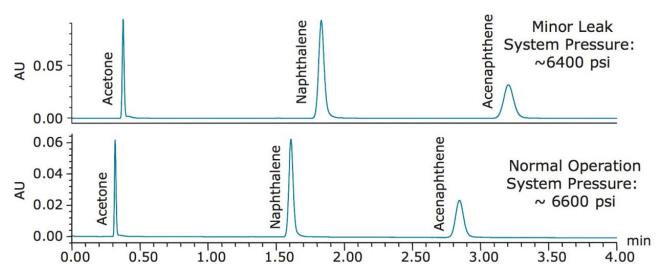


Figure 2. Separation of the Neutrals QCRM comparing a malfunctioning pump with a minor leak to a normal operating pump.

When a minor leak was present in the pump, all of the peaks were still eluting within the sample run time; however, there is a shift in retention times and a slight change in the system pressure. With stringent specifications set in the laboratory, the approximate 10% difference in retention times, shown in Table 4, may fall outside of the specifications, alerting the analyst of a potential system issue. Combined with the pressure difference, this may indicate to the analyst that the pump could be malfunctioning. After the pump was repaired, the system was re-checked for performance using the Neutrals QCRM. Nine injections of the standard were performed and the data was compiled, shown in Table 4. The combined retention time %RSDs were less than 0.7 for all peaks after the leak was repaired, which confirms that the system was back to normal operation.

	Acetone	Naphthalene	Acenaphthene
Retention Time (min)	0.376	1.832	3.206
USP Tailing Factor	1.160	1.140	1.080
USP Plate Count	3438	9348	8516
Retention Time Average (min)	0.323	1.631	2.893
Retention Time %RSD	0.680	0.450	0.440
USP Tailing Factor	1.180	1.120	1.070
USP Plate Count	3148	11020	10421
	USP Tailing Factor USP Plate Count Retention Time Average (min) Retention Time %RSD USP Tailing Factor	Retention Time (min) 0.376 USP Tailing Factor 1.160 USP Plate Count 3438 Retention Time Average (min) 0.323 Retention Time %RSD 0.680 USP Tailing Factor 1.180	Retention Time (min) 0.376 1.832 USP Tailing Factor 1.160 1.140 USP Plate Count 3438 9348 Retention Time Average (min) 0.323 1.631 Retention Time %RSD 0.680 0.450 USP Tailing Factor 1.180 1.120

Table 4. Data collected for a malfunctioning pump with a minor leak and the repaired pump (n=9).

A third common system problem is a bad check valve. Check valves help to regulate flow and pressure in an HPLC/UPLC system. Over time, these valves may stick and become clogged depending on the types of

mobile phase used. When they begin to fail, there can be noticeable chromatographic and pressure issues in a system. The separation of the Neutrals QCRM on a system with a bad check valve compared to a good check valve is shown in Figure 3.

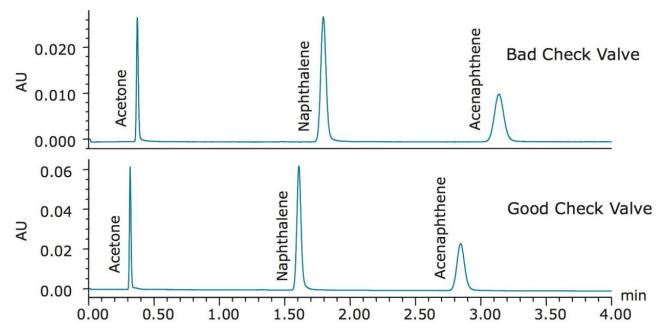


Figure 3. Separation of the Neutrals QCRM on a system with a bad check valve and good check valve. The retention times of all three peaks shift with the bad check valve compared to the good check valve. This slight change in retention time is caused by the check valve not being able to regulate the flow of the mobile phase effectively. In this instance, not only did the retention of the compounds increase, but the plate count dropped by 26% for naphthalene, shown in Table 5. This shift in retention time, as well as the decrease in plate count, may cause a run of standard to fall out of specifications. Once the check valve was replaced, the system was checked for performance once again by running an additional nine injections of Neutrals QCRM, shown in Table 5. In this instance, the plate counts and retention times of the nine injections were comparable to the benchmarked data, indicating a normally functioning system after repair of the check valve.

		A M. Lil. 1 A I		A
		Acetone	Naphthalene	Acenaphthene
Bad Check Valve (Observed during problem)	Retention Time (min)	0.374	1.805	3.156
	USP Tailing Factor	1.170	1.070	1.020
	USP Plate Count	3103	8768	8783
Bad Check Valve (After repairs)	Retention Time Average (min)	0.323	1.631	2.893
	Retention Time %RSD	0.680	0.470	0.440
	USP Tailing Factor	1.160	1.140	1.070
	USP Plate Count	3157	11025	10511

Table 5. Data collected from the system with a bad check valve and a good check valve (n=9).

A fourth common mode of failure in an LC system is improper column connection. Improperly connecting the tubing to the column can occur when changing columns and can result in a gap between tubing and column end fittings. This gap can affect peak shape, potentially widening peaks, resulting in excessive peak tailing or shouldering. The separation of the Neutrals QCRM on a column with an improper connection compared to a column that is properly connected can be seen in Figure 4.

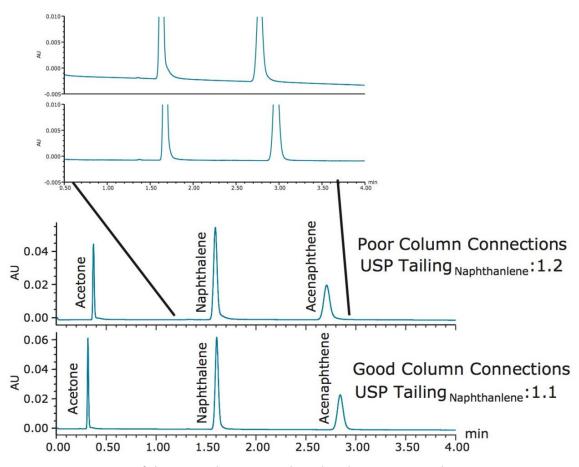


Figure 4. Separation of the Neutrals QCRM with and without proper column connections.

In this example, the separation of the Neutrals QCRM with the improper column connection shows only slight changes in the separation compared to the proper column connection. The effect of improper column connections can vary depending on the extent of the gap created. In this case, tailing of the naphthalene peak increases slightly with the poor tubing installation as well as the decreased retention of the acenaphthene peak. These changes could indicate many problems with the system. Each peak is affected differently and as the differences are only slight, they might go unnoticed. In addition to the higher tailing of the naphthalene peak, a drop in plate count is also observed, shown in Table 6.

Depending on assays and specifications, this drop in plate count may cause system performance checks to fail. By connecting the column properly and injecting nine injections of the standard, the system performance was re-checked, shown in Table 6. The tailing factor for naphthalene returned to approximately 1.1, and the plate counts increased and returned to the same performance as the benchmarked data, indicating that the system has returned to optimal operation.

		Acetone	Naphthalene	Acenaphthene
	Retention Time (min)	0.385	1.669	2.815
Improper Seating of Column Outlet Tubing (Observed during problem)	USP Tailing Factor	1.140	1.200	1.170
(Observed during problem)	USP Plate Count	3159	8685	8659
Improper Seating of Column Outlet Tubing	Retention Time Average (min)	0.323	1.631	2.893
	Retention Time %RSD	0.670	0.450	0.440
(After repairs)	USP Tailing Factor	1.160	1.080	1.070
	USP Plate Count	3220	10980	10438

Table 6. Data collected with improper and proper column connections (n=9).

A fifth common problem in LC is an air bubble in the solvent line, which can be caused by inadequate system priming or running out of solvent in the solvent bottles. Once an air bubble forms, it can affect the system pressure and mobile phase delivery. The effect of an air bubble on the separation of the Neutrals QCRM is shown in Figure 5.

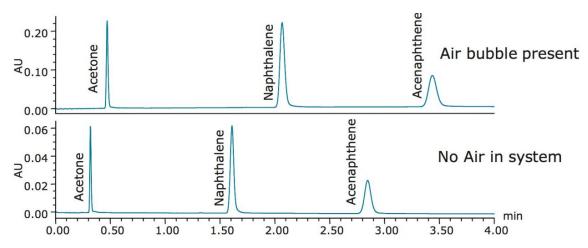


Figure 5. Separation of the Neutrals QCRM with and without an air bubble in the system.

With air in the system, the retention time of all the peaks has shifted. The air in the solvent lines or pump can cause improper delivery of the mobile phase, thereby shifting retention time. In this case, a 25% increase in retention of the naphthalene peak was observed, shown in Table 7. By re-priming the system with mobile phase, the air was removed from the system. Looking at the data from nine injections of the Neutrals QCRM after removing air from the system, shown in Table 7, the retention times have returned to where they were during system benchmarking, indicating the system no longer has an air bubble.

		Acetone	Naphthalene	Acenaphthene
Air Bubble in system (Observed during problem)	Retention Time (min)	0.471	2.066	3.436
	USP Tailing Factor	1.120	1.160	1.110
(Observed during problem)	USP Plate Count	4408	10116	9616
Air Bubble in system (After repairs)	Retention Time Average (min)	0.323	1.631	2.893
	Retention Time %RSD	0.690	0.460	0.440
	USP Tailing Factor	1.15	1.06	1.09
	USP Plate Count	3189	11202	10449

Table 7. Data collected with and without an air bubble in the LC system. The air bubble was removed by priming the mobile phase pump.

The final common problem seen in LC that was studied in this application is varying organic composition in the mobile phase, which can happen during mobile phase preparation due to analyst error. Slight variations in mobile phase composition can have effects on chromatographic results, including increasing or decreasing retention times and potentially causing co-elution of peaks. In this application, the percentage of acetonitrile was altered by \pm 2% for the analysis of the Neutrals QCRM. Figure 6 shows the separation with using mobile phase compositions of 48%, 52%, and 50% (recommended composition) acetonitrile.

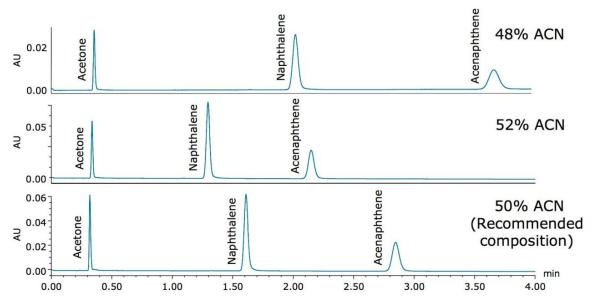


Figure 6. Impact of mobile phase strength on the separation of the Neutrals QCRM.

Predictably, the varying organic composition has a significant effect on the isocratic separation of the standard. A retention time shift of 25% for the naphthalene peak was observed when the mobile phase contained 48% acetonitrile, while a 21% decrease in retention time occurred when 52% acetonitrile was present in the mobile phase, shown in Table 8. Once the original mobile phase composition (50% acetonitrile) was placed back onto the system, nine injections of the Neutrals QCRM were run to recheck the system performance, shown in Table 8, and to demonstrate proper system operation. The retention times of all the peaks were comparable to the benchmarked data, indicating that the mobile phase was made accurately and that the system is functioning as it should. The Neutrals QCRM is compatible with many mobile phases, and if the mobile phases for sample analysis are used to both benchmark and troubleshoot the system, the benefits of the Neutrals QCRM for troubleshooting mobile phase errors can be realized. While this application focuses on the use of 50% acetonitrile, other mobile phases with sufficient organic composition may be used. Without the use of a standard to check system performance, an error in mobile phase preparation could cause irreproducible chromatography or coelution of target peaks in real samples, resulting in extensive and unnecessary method development, or reanalysis of the samples. Instead, with proper specifications for the Neutrals QCRM, errors in organic composition may be identified before time is invested in sample analysis.

		Acetone	Naphthalene	Acenaphthene
52% ACN used in separation	Retention Time (min)	0.336	1.295	2.148
	USP Tailing Factor	1.130	1.070	1.080
	USP Plate Count	3290	9323	9144
48% ACN used in separation	Retention Time (min)	0.358	2.035	3.690
	USP Tailing Factor	1.150	1.080	1.080
	USP Plate Count	3124	8818	8974
	Retention Time Average (min)	0.323	1.632	2.893
50% ACN (recommended conditions) used in separation	Retention Time %RSD	0.680	0.430	0.440
	USP Tailing Factor	1.140	1.050	1.060
	USP Plate Count	3297	11402	10520

Table 8. Data collected from the injections of Neutrals QCRM with different mobile phase compositions.

Conclusion

The Waters Neutrals QC Reference Material (QCRM) is a mixture of three neutral compounds that can be separated with common mobile phases with sufficient organic composition, and is compatible with most column chemistries, making it an ideal standard for troubleshooting HPLC and UPLC system problems. Before the standard can be used as a troubleshooting tool, it is recommended to benchmark the system performance using the standard and create a set of specifications to determine the limits of acceptable data for future runs of the standard. During routine analysis the standard can be used to monitor to the system and if a problem arises, the Neutrals QCRM can be run to determine if a system problem exists and to help identify the issue. Once the problem is resolved, the standard can be run again to confirm that the system is back to normal operation.

There are many benefits to using the Neutrals QCRM. First, system downtime can be reduced. This allows for a better use of resources, with more samples analyzed and less time spent by analysts trying to fix a system problem. Secondly, as a result of the strict manufacturing process of the standard, errors in suitability standard preparation are mitigated, isolating issues to the system and allowing a chromatographer to identify the problem faster. Lastly, the Neutrals QCRM can be used to ensure that a recently repaired system is functioning optimally. This increases the confidence in the data produced after repairs and ensures high quality data generation.

References

- 1. Berthelette KD, Summers M, Fountain KJ. Benchmarking System Performance using Waters Neutrals QC Reference Material. Waters Application Note Literature code: 720004622EN
- 2. Quality Control Reference Material and Benchmarking Instrument Performance. Waters White Paper Literature code: 720004535EN.

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