

應用手冊

Impurity Profiling Using UPC²/MS

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Abstract

In this application note, we use the ACQUITY UPC² System coupled to ACQUITY SQD to analyze the identity and relationship of the unknown peaks observed during the method development standards and expired samples of metoclopramide.

Benefits

- · Solve complex impurity profile challenges
- · Approaches to aid final method development decisions

Introduction

UltraPerformance Convergence Chromatography (UPC²) exploits the benefits of sub-2-µm particle size stationary phases, with carbon dioxide as the primary mobile phase component. Convergence chromatography is a complementary analytical technique to liquid chromatography as it provides orthogonal selectivity, thereby increasing the opportunity to identify impurities present in a sample. Mass spectral information helps analysts confirm, identify, and characterize the quality of the pharmaceutical ingredients. Coupling UPC² to mass spectrometry provides an important tool for pharmaceutical analysis compared to previously published reversed phase liquid chromatography (RPLC) impurity analysis approaches.¹⁻³

Anomalies were observed during the method development screening process.⁴ In one instance, a standard solution of impurity F was hypothesized to be unstable after a few days. In this application, we use the ACQUITY UPC² System coupled to ACQUITY SQD to analyze the identity and relationship of the unknown peaks observed during the method development standards and expired samples of metoclopramide. Impurity relationship to the API are hypothesized and confirmed with the use of the MS spectral data. Finally, the MS data from the impurity profile was interrogated to ensure the specificity of the methodology in the presence of these unknown peaks to aid future refinement of the final method.

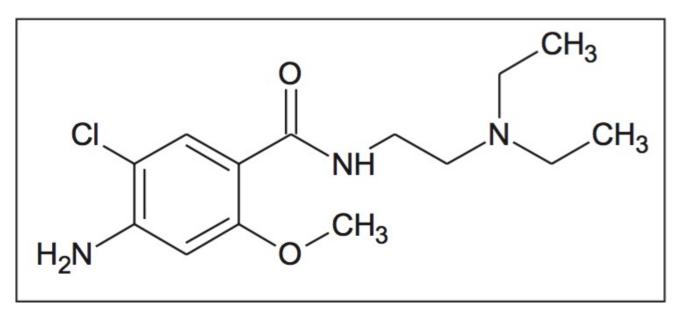


Figure 1. Chemical structure of metoclopramide.

Experimental

UPC² conditions

System:	ACQUITY UPC ² with PDA and SQD detection
Column:	ACQUITY UPC ² BEH 2-EP 3.0 x 100 mm, 1.7 μm
Mobile phase A:	CO ₂
Mobile phase B:	1 g/L Ammonium formate in 50:50 methanol/acetonitrile spiked with 3% formic acid
Wash solvents:	70:30 methanol/ isopropanol
Separation mode:	Gradient; 5% to 30% B over 5.0 min; held at 30% for 1 min
Flow rate:	2.0 mL/min

UPC² conditions

CCM back pressure: 1500 psi

Column temp.: 50 °C

Sample temp.: 10 °C

Injection volume: $0.5 \mu L$

Run time: 6.0 min

Detection: PDA 3D Channel: PDA, 200 to 410 nm; 20Hz

PDA 2D Channel: 275 nm at 4.8 nm Resolution

(compensated 500 to 600 nm)

SQD MS: 150 to 1200 Da; ES PosNeg

Make-up flow: None; (make-up pump not configured in flow

splitter)

Data management: Empower 3 Software

Sample description

Investigations of impurity C and impurity F standards were prepared in methanol and explored at 0.1 mg/mL concentrations. The expired sample preparation was extracted using methanol, and prepared at a concentration of 2 mg/mL relative to the metoclopramide active ingredient.

Results and Discussion

Prior to analyzing the degraded metoclopramide sample, the MS data was examined to address questions regarding observations with the chromatography of impurity reference standards 4-amino-5-chloro-2-methoxybenzoic acid; "impurity C" and 4-amino-5-chloro N-2-(diethylaminoethylo-2-hydroxybenzamide

MS investigation of reference standard impurity C

During the screening process, two peaks were separated during the injection of the impurity C standard using the ACQUITY UPC² CSH Fluoro-Phenyl column. This phenomenon was not observed with the other stationary phases. MS spectral analysis of the two peaks showed similar spectra. The MS spectrum indicates possible dimerization of the analyte. Based on this information, it was determined that the two peaks were related to each other rather than the second peak being a contaminant. The rapid determination of this relationship provided direction for scoping future analysis by MS/MS and accurate mass. The data generated by those techniques will be more useful in determining the purity of the standard.

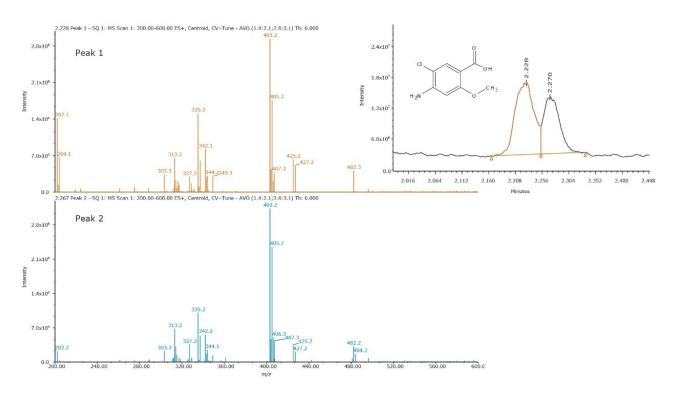


Figure 2. MS spectral analysis of EP impurity C for the doublet peaks observed when using the UPC² CSH Fluoro-Phenyl stationary phase.

Investigation of impurity F working standard solution stability

The peak shape of impurity F was observed to degrade over time, during the method development process.

Unknown peaks would appear in the chromatogram within the course of a week. In addition, the color of the standard solution changed from a clear solution to a solution with a brownish tint. MS interrogation revealed the masses listed in Table 1. The masses were correlated to those found to be significant to the UV trace (not shown)

between retention time 2.0 min and 3.5 min. XIC of m/z = 330 and 296 resulted in multiple peaks.

Rt (min)	Observed m/z	Δ Mass	Δ Mass Proposed transformation	
2.924	286			
2.268	344	+ 58 Da	methoxylation + methylation	
2.303 & 2.614	330	+ 44 Da	methoxylation	
2.680 & 2.886	296	+ 10 Da	hydrolysis + two methylations	
2.864	356	+ 70 Da	?	
3.113	252	- 34 Da	Loss of Cl-	
3.288	258	- 28 Da	Loss of two CH ₃ groups	
	2.924 2.268 2.303 & 2.614 2.680 & 2.886 2.864 3.113	2.924 286 2.268 344 2.303 & 2.614 330 2.680 & 2.886 296 2.864 356 3.113 252	2.924 286 2.268 344 + 58 Da 2.303 & 2.614 330 + 44 Da 2.680 & 2.886 296 + 10 Da 2.864 356 + 70 Da 3.113 252 - 34 Da	

Table 1. Masses found in the degraded standard solution of metoclopramide EP impurity F.

The working standard was prepared in methanol. Many of the impurity peaks were products of methylation or methoxylation. Based on this information, alternative diluents will be explored to inhibit the likelihood of these transformations. Presently, the working standard solution shelf life has been decreased to three days until a suitable diluent study can be performed. The unknown peak #8 in Table 1 with m/z = 258 has the same mass of methyl 4-(acetylamino)-5-chloro-2-methoxybenzoate, commonly referred to as "impurity B." This unknown peak found in the impurity F working standard was determined not to be identical to EP impurity B due to differences in retention time; whereas, impurity B elutes at approximately 0.48 min. The origins of other impurities are still undetermined. They are hypothesized to be more products of solution instability or present in the reference material.

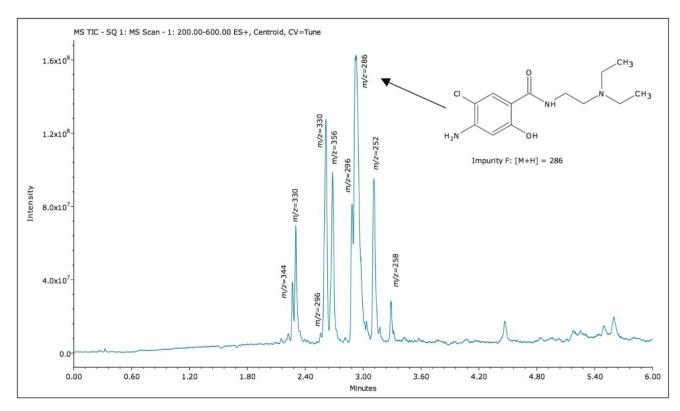


Figure 3. MS ES+ TIC of a degraded standard solution of metoclopramide EP impurity F.

Investigating an expired metoclopramide sample

The ACQUITY UPC² System coupled to the ACQUITY SQD Mass Spectrometer controlled by Empower 3 Software provided a simple solution to profile impurities in an expired metoclopramide sample. MS spectral extraction of peaks in the UV chromatographic trace were simply performed by using right mouse click in the review window to rapidly confirm known impurities and identify 12 unknown impurities, as shown in Table 2. The sensitivity of UPC² provided $s/n \ge 10$ for impurities detected with area% $\ge 0.05\%$ in the UV chromatographic trace. The expired metoclopramide sample was interrogated to determine if the masses in Table 1 were present. The MS data confirmed the presence of 4 out of the 8 impurities related to EP F; m/z = 296, 344, 252, and 258. In addition to the known impurities, a total of 7 unknown impurities were detected. The masses, retention time, and UV signal-to-noise were recorded in Table 2.

	Rt	% Area	USP s/n	m/z
Impurity B	0.49	0.48	148	258
Unknown #1	0.90	0.34	54	226
Impurity "9"	1.18	1.34	255	(ESI-) 180
Impurity D	1.41	1.14	229	224
Unknown #2	2.18	0.12	22	306
Unknown #3	2.33	0.02	8	358
Unknown #4	2.38	1.29	195	n/a
Impurity A (Isomer 1)	2.49	0.28	48	342
Impurity A (Isomer 2)	2.53	0.41	65	342
Rel. Impurity F (Unk. #1)	2.61	0.69	87	344
Rel. Impurity F (Unk. #6)	2.83	0.06	10	296
Impurity F	2.91	0.51	82	286
Rel. Impurity F (Unk. #7)	3.09	0.14	20	252
Unknown #5	3.14	0.55	77	310
Rel. Impurity F (Unk. #8)	3.28	0.14	20	258
Impurity G	3.37	0.87	130	316
Metoclopramide	3.43	86.07	8186	300
Unknown #6	3.60	2.83	380	266
Unknown #7	3.63	1.09	165	272
Impurity H	4.53	1.62	101	(ESI-) 194

Table 2. Peak list of the UV 275 nm chromatographic trace relative to the injection of the metoclopramide expired sample.

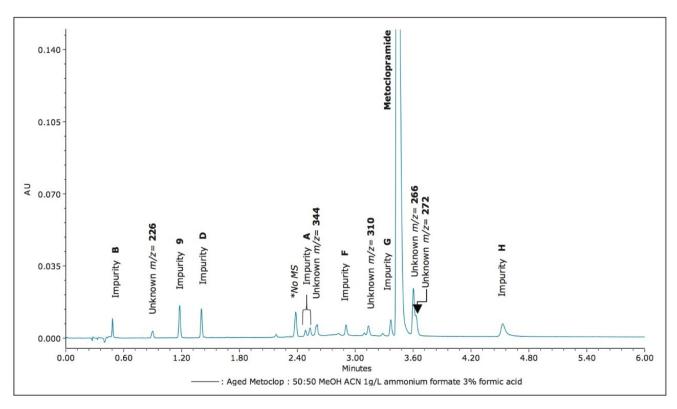


Figure 4. UV 275 nm chromatographic trace of a degraded sample of metoclopramide. The major impurities are labeled.

Conclusion

The ACQUITY UPC² System coupled to MS provided a comprehensive approach to impurity profiling. This configuration enables a scientist to quickly answer questions regarding reference standard purity, as shown with reference standard impurity C. UPC²-MS guided the decisions to investigate diluent choices for the impurity F working standard and adjusting the shelf life of the working standard solution. In addition, investigating instability of impurity F provided insight into other potential impurities that may be present in the drug sample impurity profile. Furthermore, seven unknown impurities were detected in the expired metoclopramide sample. Interrogation of the UV and MS data was simply performed using Empower 3 Software. Overall, utilizing UPC²-MS increased the knowledge base of pharmaceutical product quality, and improved the methodology procedures involved with achieving the analytical goals.

References

- 1. Ahuja S, Alsante K. Handbook of Isolation and Characterization of Impurities in Pharmaceutical Compounds. *Elsevier*. 2003.
- 2. Jones MD et al. Identification and Characterization of an Isolated Impurity Fraction: Analysis of an Unknown Degradant Found in Quetiapine Fumarate. Waters Application Note 720003079EN. 2009 Oct.
- 3. Jones MD et al. A Workflow Approach for the Identification and Structural Elucidation of Impurities of Quetiapine Hemifumarate Drug Substance. Waters Application Note 720003081EN. 2009 Oct.
- 4. Jones MD, Potts W. UPC² Method Development for Achiral Impurity Analysis. Waters Application Note 720004577. 2013 Jan.

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