

# Oasis 2x4 Method: Proof of Concept

Waters Corporation



This is an Application Brief and does not contain a detailed Experimental section.

## Abstract

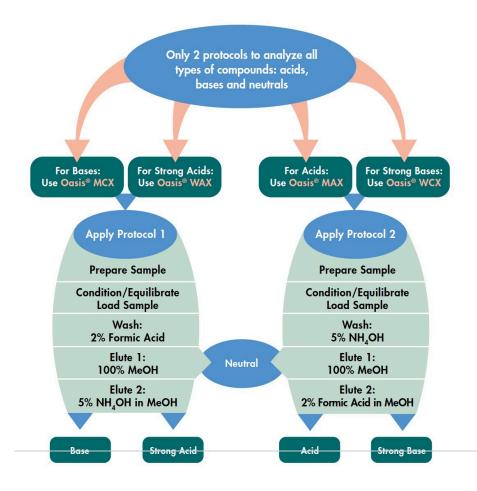
This application brief highlights Oasis 2x4 Method which is a viable approach to SPE sorbent and protocol selection.

#### Introduction

In order to prove that the Oasis 2x4 Method is a viable approach to SPE sorbent and protocol selection, a group of representative small molecules was spiked into rat plasma. The molecules include a base, a quaternary ammonium salt, a neutral, an acid and a strong acid. The spiked plasma samples were then extracted following the Oasis 2x4 Method. All molecules elute in the correct Elute 1 or Elute 2 fractions according to what theory predicted, proving that the method is viable.

Follow the simple steps outlined in this flow chart to achieve high recoveries and the cleanest extracts:

- · Characterize your analyte [Neutral, Acid or Base, pKa].
- · Select one of the four Oasis sorbents.
- · Apply the indicated Protocol [1 or 2].
- · Determine SPE recoveries by LC analysis.



# Experimental

10 mg Oasis 96-well Plates

### Protocol 1

| Condition:   | 500 μL MeOH   |
|--------------|---|
| Equilibrate: | 500 μL H <sub>2</sub> O   |
| Load:        | 500 $\mu L$ sample (250 $\mu L$ plasma diluted 1:1 with 4% $H_3 PO_4$ in $H_2 O)$ |
| Wash 1:      | 500 $\mu L$ 2% HCOOH in $H_2O$  |
| Elution 1:   | 2 x 125 μL MeOH   |
| Elution 2:   | 2 x 125 $\mu L$ 5% $NH_4OH$ in MeOH   |
| Dilution:    | 250 μL water  |
| Protocol 2   |   |

| Elution 1:   | 2 x 125 μL MeOH   |
|--------------|---|
| Wash 1:      | 5% NH <sub>4</sub> OH in H <sub>2</sub> O   |
| Load:        | 500 $\mu L$ sample (250 $\mu L$ plasma diluted 1:1 with 4% $H_3 PO_4$ in $H_2 O)$ |
| Equilibrate: | 500 μL H <sub>2</sub> O   |
| Condition:   | 500 μL MeOH   |

| Elution 2:          | $2 \ x \ 125 \ \mu L \ 2\%$ HCOOH in MeOH                                     |
|---------------------|---|
| Dilution:           | 5% NH <sub>4</sub> OH in H <sub>2</sub> O (To neutralize acid for high pH LC) |
| Column:             | XBridge C <sub>18</sub> 2.1 x 20 mm IS, 3.5 μm                                |
| Mobile Phase A:     | 10 mM NH <sub>4</sub> HCO <sub>3</sub> , pH 10                                |
| Mobile Phase B:     | 10 mM NH4H  |
| Injection Volume:   | 10.0 μ L  |
| Column Temperature: | Ambient   |
| Detection:          | UV @ 254 nm (Prednisone)  |
| Instrumentation:    | 2777 Sample Manager, 1525µ Binary HPLC Pump,<br>Quattro Premier and 2996 PDA  |

## Gradient

| Time (min) | Profile |
|------------|---------|
|            | %A      |
| 0          | 95      |
| 3.0        | 5       |
| 4.8        | 5       |
| 5.0        | 95      |

| Time (min)               | Profile                                |
|--------------------------|--|
| 7.0                      | 95                                     |
|                          |  |
| Quattro Premier          |  |
| Capillary:               | 3.4 kV                                 |
| Source Temp.:            | 120 °C                                 |
| Desolvation Temp.:       | 350 °C                                 |
| Cone Gas Flow:           | 50 L /Hr                               |
| Desolvation Gas Flow:    | 700 L /Hr                              |
| Collision Cell Pressure: | 2.59e <sup>-3</sup> mbar               |
| MRM Transitions:         | Imipramine 281.2 > 85.95 ESI+          |
|                          | Decanesulfonic Acid 220.9 > 79.7 ESI - |
|                          | lbuprofen 205.2 > 161 ESI -            |
|                          | Valethamate 306.3 > 162.8 ESI+         |

# Results and Discussion



Imipramine (B) pKa = 9.4 100 ng/mL

Prednisone (N) 20 µg/mL

Valethamate (QA) pKa >12 100 ng/mL

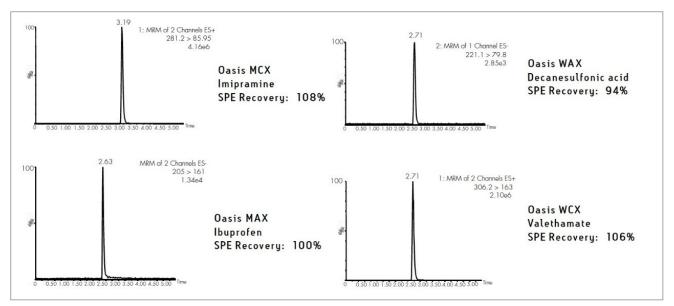
0 \$`0- Na⁺

Decanesulfonic Acid (SA) pKa <0.5 200 ng/mL

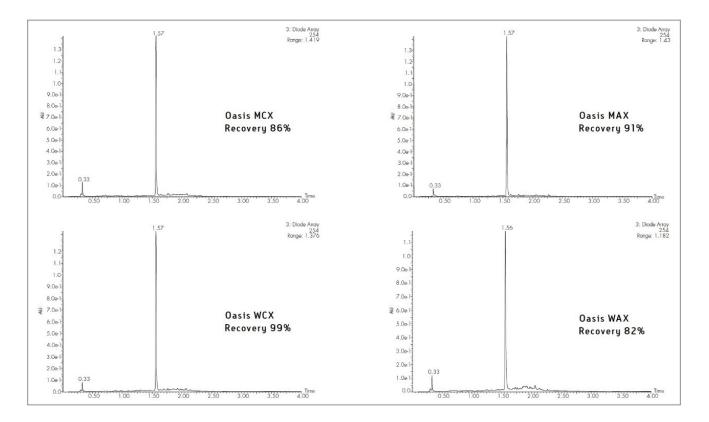
OH

lbuprofen (A) pKa = 5.2 100 ng/mL

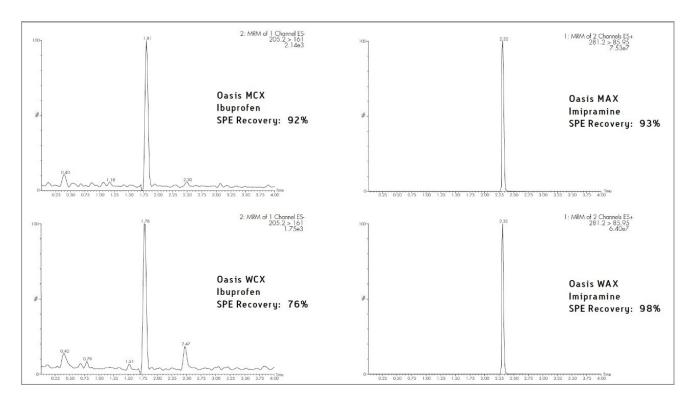
#### Elute 2, Primary Analyte Data



Prednisone SPE Recovery Data



#### Elute 1, Counter Analyte Data



## Featured Products

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WA60090, June 2007

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