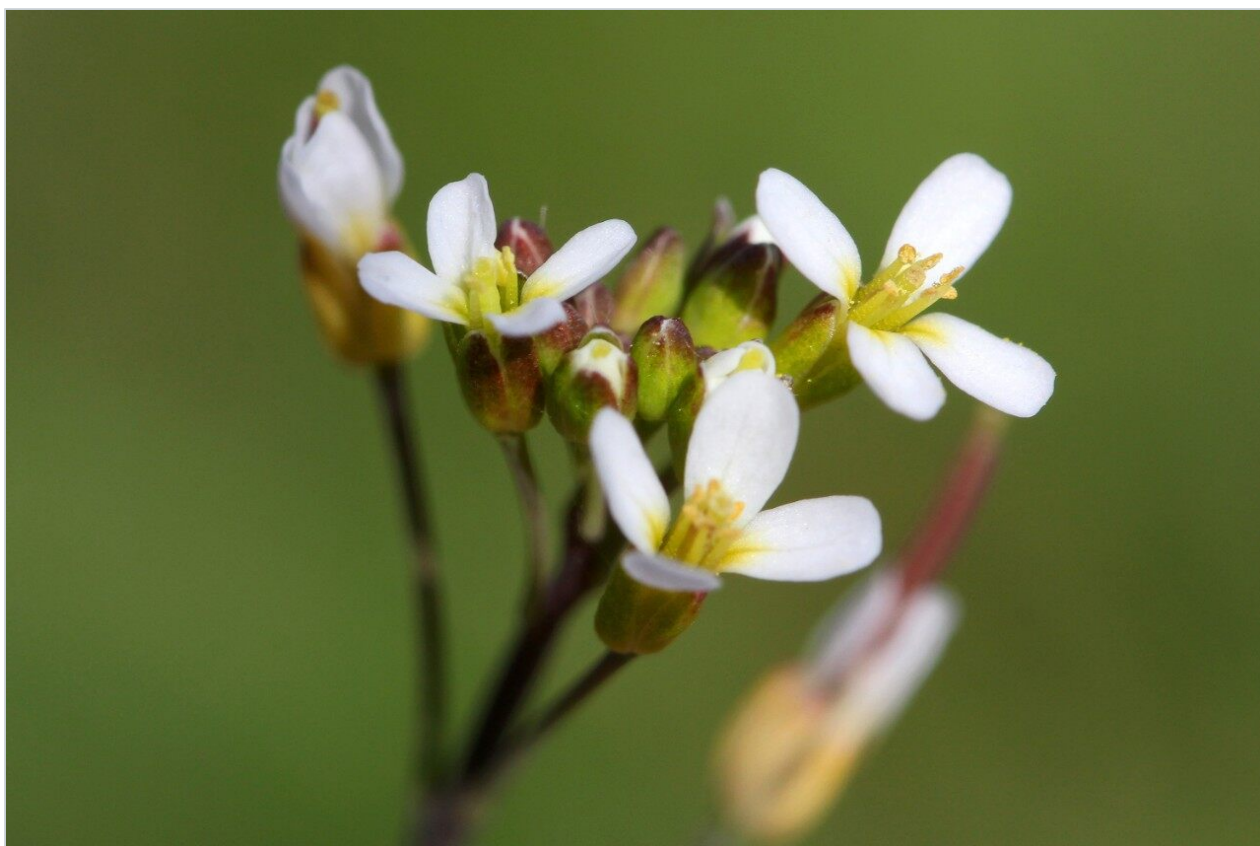


응용 자료

An Exact Mass Metabolomic Study of Nitrogen Starvation in *Arabidopsis thaliana* using the GCT Premier

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Abstract

In this application note, we use GC-MS with the Waters GCT Premier mass spectrometer to study nitrogen starvation in *Arabidopsis thaliana*, focusing on the metabolic profiles of roots and leaves after moderate and severe nitrogen starvation.

Introduction

Nitrogen is an essential element for plant growth and development. In natural soils, nitrogen is often a significant factor limiting plant growth. Nitrogen stress triggers various responses at the level of metabolism, gene expression and development allowing the sessile plant to adapt by short and long term mechanisms. Nitrogen assimilated in biomolecules can be released back to inorganic nitrogen (NH_4), which can re-enter metabolism in various physiological processes such as photorespiration and biosynthesis of phenylpropanoids. Due to the coordination of nitrogen metabolism with other pathways, e.g. carbon metabolism, major changes in carbon and nitrogen metabolism have been described (Scheible et al. 1997),¹ but more detailed analysis of metabolite profiles are needed to understand the dynamic response of the metabolic network to nitrogen stress. Recent studies on tomato plants (Urbanczyk-Wochniak and Fernie 2005)² and *Chlamydomonas* (Bolling and Fiehn 2005)³ have demonstrated the immense changes in metabolite profiles after nitrogen starvation.

In this application note, we use GC-MS with the Waters GCT Premier Mass Spectrometer to study nitrogen starvation in *Arabidopsis thaliana*, focusing on the metabolic profiles of roots and leaves after moderate and severe nitrogen starvation. Moderate starvation was studied two days after nitrogen withdrawal and severe starvation after 10 days. A detailed comparison of the metabolic profiles of these two organs over a time course of starvation will give further understanding of the adaptive behavior of plants to nitrogen stress.



Figure 1. Waters GCT Premier.

Experimental

Plant growth and sample preparation *Arabidopsis thaliana* plants were cultivated under hydroponic conditions using the method of Orsel et al.⁴ The plants were grown for five weeks on 6 mM nitrate in a hydroponic device using short days. Total nitrogen starvation was applied for either two days or 10 days and roots and leaves were sampled separately.

40 mg of *Arabidopsis* leaves and roots (fresh material) were extracted with 1 mL extraction buffer (Gulberg et al)⁵. The extraction buffer (chloroform/methanol/water (1:3:1 V/V/V at -20 °C) allowed the extraction of lipophilic and hydrophilic metabolites in one phase. All samples were vortexed for 3 minutes and then centrifuged for 10 minutes at 3000 rpm and 4 °C. 300 µL of supernatant was dried under vacuum and the dried pellets were conserved under argon and stored at -80 °C prior to analysis.

The dried extracts were derivatized using a two-stage process based on the method of Fiehn et al.⁶ 20 µL of 40 mg/mL methoxyamine hydrochloride in pyridine was added to the dried extracts and held at 28 °C for 90

minutes. This was followed by the addition of 180 μ L of MSTFA for 30 minutes at 37 °C. The samples were analyzed by GC-MS as follows:

GC Conditions

GC system:	Agilent 6890N
Column:	J&W Scientific DB-5MS 30 m x 0.25 mm i.d. x 0.25 μ m film
Flow rate:	1.0 mL/min helium
Transfer line:	280 °C
Solvent delay:	4.3 min

Gradient

Temperature(°C)	Time(min)	Rate(°C/min)
85	2	15
320	5	-

MS Conditions

MS system:	Waters GCT Premier
Ionization mode:	El and CI+
Source temperature:	200 °C
Electron energy:	70 eV
Trap current (EI):	200 μ A

Emission current (CI+):	200 μ A
Mass range:	<i>m/z</i> 50 to 1000
Acquisition rate:	0.19 sec
CI Reagent gas:	90:10 methane/ammonia
Lock reference EI:	Chloropentafluorobenzene
Exact mass:	201.9609
Lock reference CI:	2,4,6-Tris(trifluoro methyl)-1,3,5-triazine
Exact mass:	286.0027

Results and Discussion

Acquired data was processed through the Waters MarkerLynx Application Manager for MassLynx software and the resulting PCA scores plot from all of the EI data is shown in Figure 2. This shows clear separation of leaves and roots and the samples from each stage of the nitrogen starvation experiment.

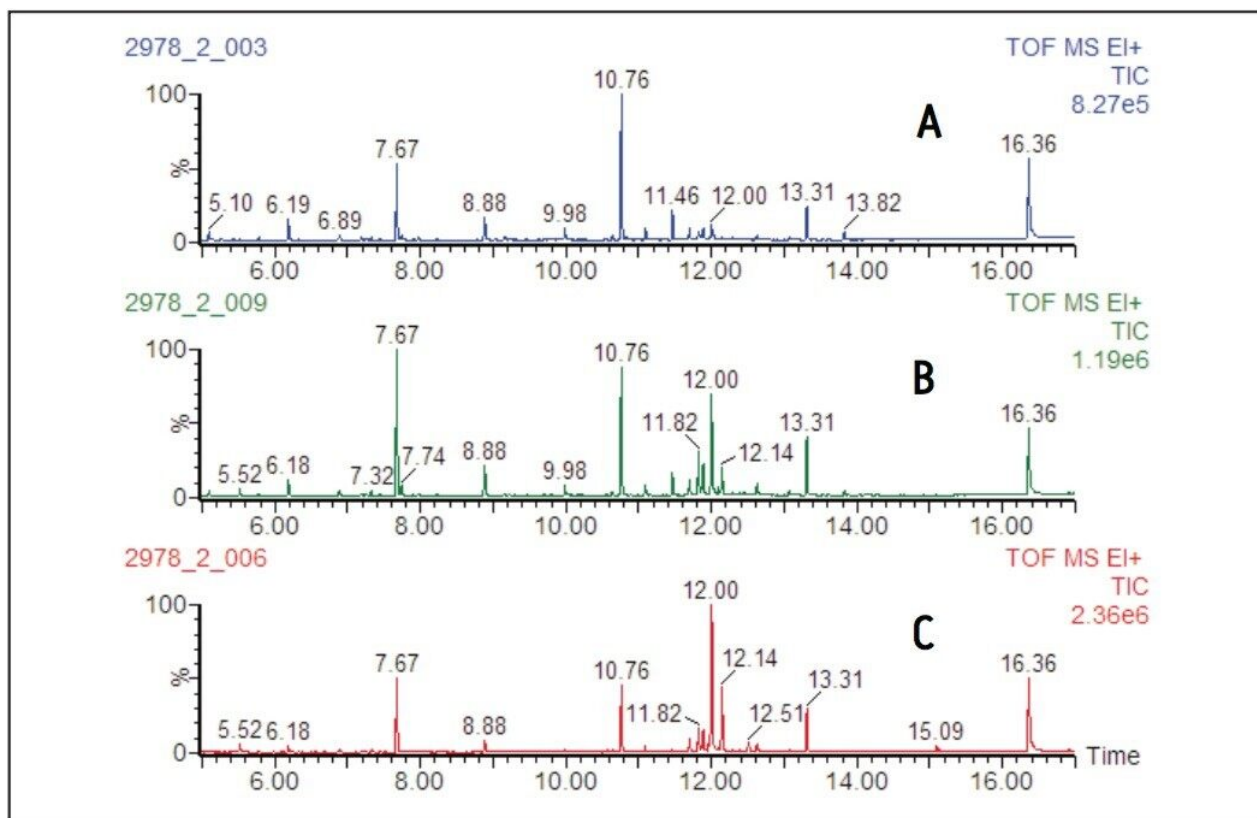


Figure 3. Representative EI TIC chromatograms from leaf extracts from (a) unstarved, (b) after two days and (c) after 10 days nitrogen starvation.

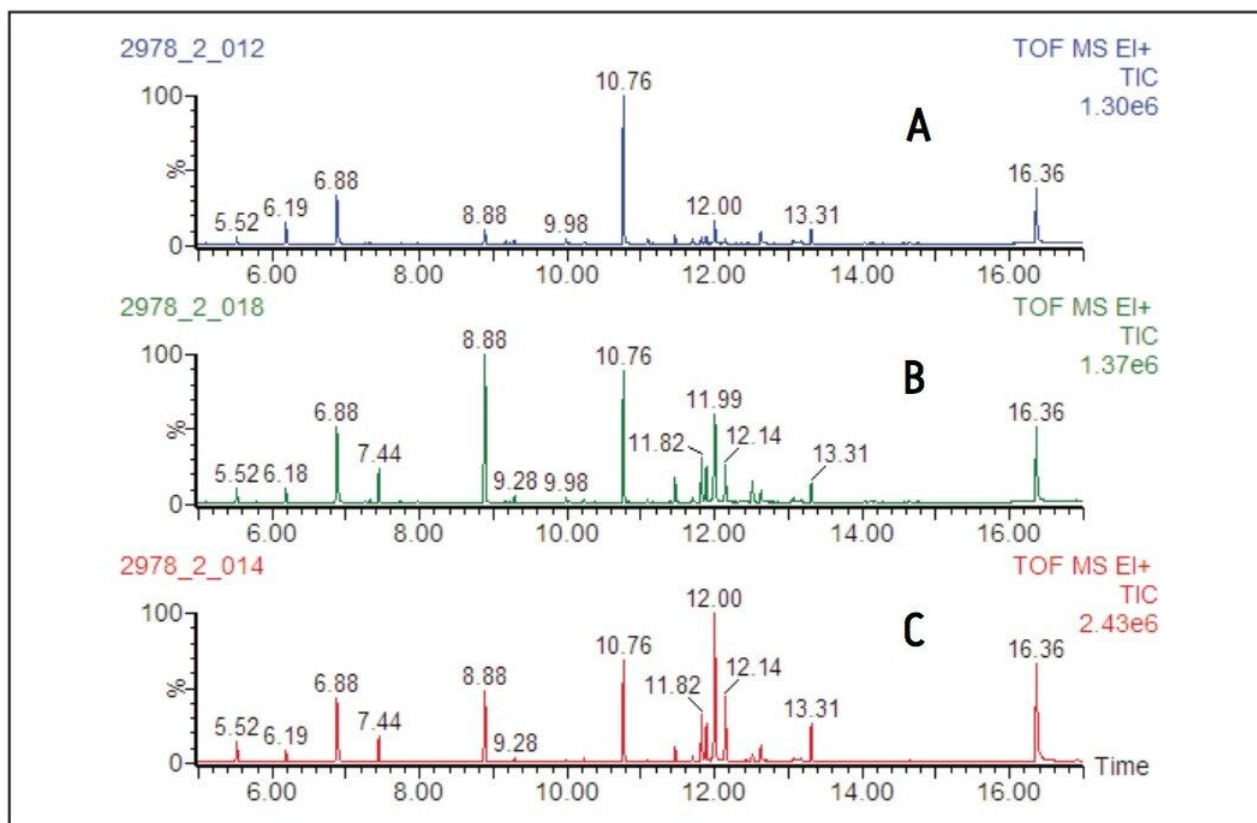


Figure 4. Representative EI TIC chromatograms from root extracts from (a) unstarved, (b) after two days and (c) after 10 days nitrogen starvation.

Examination of the chromatograms shows distinct differences between the metabolic profiles of the leaves and roots, with the major peak at 7.67 minutes in the leaf samples, assigned as fumaric acid, only being present at trace levels in the root samples.

Conversely, in the root samples the peak at 6.88 minutes, mainly composed of tris(trimethylsilyl) phosphate, is only present at a low level in the leaf samples.

The major changes observed as a result of the nitrogen starvation, however, were an increase in intensity of the peaks in the retention time range of 11.75 to 12.25 minutes, corresponding to an increase in carbohydrates.

The levels of amino acids were observed to decrease during the time course of the nitrogen starvation, consistent with the results reported by Urbanczyk-Wochniak and Fernie.² The amino acids were identified using the NIST library of EI spectra and their molecular masses confirmed by exact mass CI analysis.

Representative EI and CI spectra obtained from a leaf sample grown under conditions of adequate nitrogen

for plant growth are shown in Figure 5. The EI spectrum gave a good match for the bis-trimethylsilyl derivative of alanine and the molecular mass was confirmed by exact mass CI with a measured mass of 234.1345 (error -0.1 mDa, -0.4 ppm) for the protonated molecule.

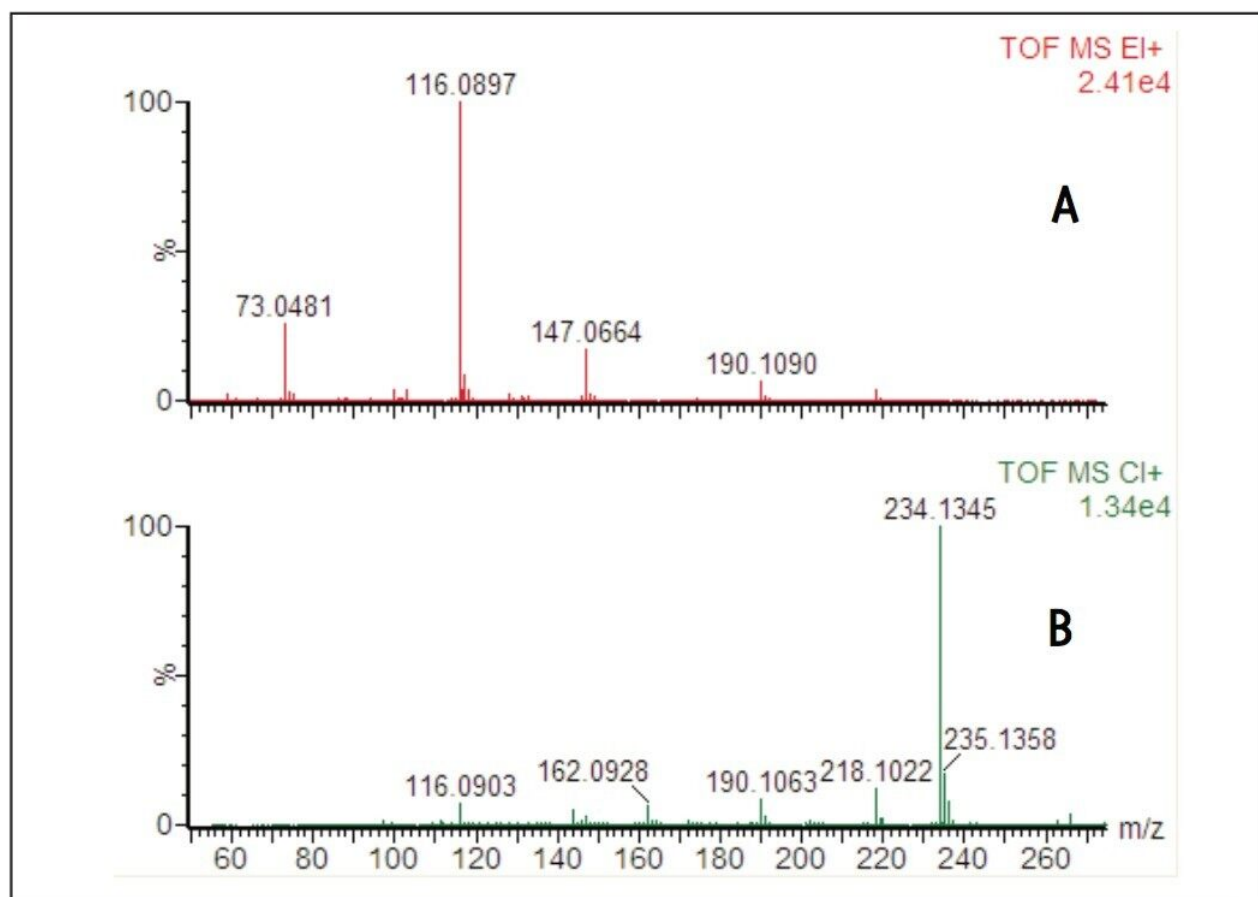


Figure 5. (a) EI and (b) CI spectra from peak eluting at 5.09 minutes from leaf extract, grown under adequate nitrogen, identified as *N,O*-Bis-(trimethylsilyl) alanine.

Metabolite identification by GC-MS is routinely carried out by comparing acquired EI spectra to spectral libraries such as NIST. However, many plant metabolites are either not in the library or the EI spectra are dominated by the derivatized groups, making de novo identification of unknown peaks difficult. CI is a less energetic process and often results in the formation of molecular ion species with reduced fragmentation, allowing access to molecular ion information.

Table 1 lists the calculated masses of the protonated molecules of some of the amino acids showing a significant reduction in both roots and leaves after nitrogen starvation and the masses reported after processing all the CI data through MarkerLynx. It can be seen that exact mass measurements of <5 ppm can

be obtained routinely giving confidence in the identification of the elemental compositions of any unknown metabolites. This confidence is further enhanced by the use of i-FIT for isotope matching.

RT (min)	Identity from NIST Library	Elemental Composition	Calculated [M+H] ⁺	Measured [M+H] ⁺	Error (mDa)	Error (ppm)
5.09	alanine	C ₉ H ₂₃ NO ₂ Si ₂	234.135	234.135	-0.1	-0.4
7.18	proline	C ₁₁ H ₂₅ NO ₂ Si ₂	260.150	260.151	0.7	2.7
7.26	glycine	C ₁₁ H ₂₉ NO ₂ Si ₃	292.158	292.159	0.3	1
9.22	pyroglutamate	C ₁₁ H ₂₃ NO ₃ Si ₂	274.130	274.130	0	0

Table 1. Table of results for amino acids showing a significant decrease in roots and leaves after nitrogen starvation, identified by searching against the NIST EI Database.

An example of the use of i-FIT can be seen in Figure 6 where five potential elemental compositions are reported for *m/z* 260, within the 5 ppm limit set. However, closer examination of the results shows that the smallest i-FIT (3.1) corresponds to the correct elemental composition for the trimethylsilyl derivative of proline, C₁₁H₂₆NO₂Si₂.

Single Mass Analysis

Tolerance = 5.0 PPM / DBE: min = -1.5, max = 50.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Odd and Even Electron Ions

1304 formula(e) evaluated with 5 results within limits (up to 20 best isotopic matches for each mass)

Elements Used:

C: 0-500

H: 0-1000

N: 0-10

O: 0-10

Si: 0-10

Mass	Calc. Mass	mDa	PPM	DBE	Formula	i-FIT	C	H	N	O	Si
260.1504	260.1502	0.2	0.8	1.5	C11 H26 N O2 Si2	3.1	11	26	1	2	2
	260.1503	0.1	0.4	-1.5	C4 H22 N7 O4 Si	110.9	4	22	7	4	1
	260.1511	-0.7	-2.7	7.5	C13 H18 N5 O	131.0	13	18	5	1	
	260.1498	0.6	2.3	2.5	C12 H22 N O5	142.8	12	22	1	5	
	260.1498	0.6	2.3	8.0	C11 H16 N8	149.7	11	16	8		

CF 6mM Nitrate

2978_2_062 437 (7.190) Cm (436:438-(431:432+440:442))

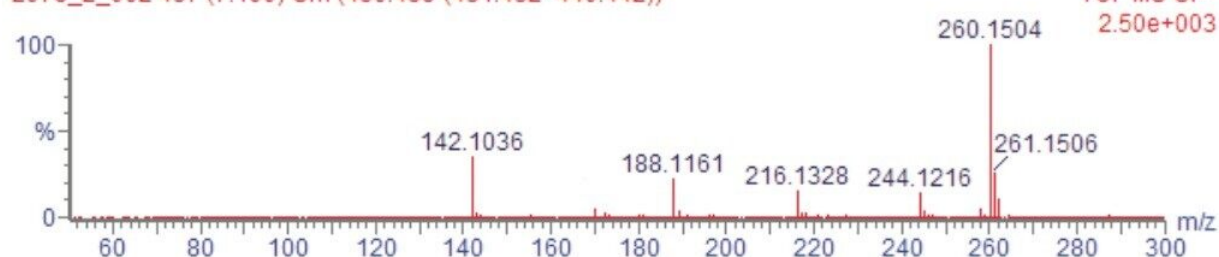


Figure 6. Elemental composition report with i-FIT for the protonated molecule of derivatized proline.

Conclusion

The combination of exact mass EI and CI analysis with processing through the MarkerLynx Application Manager has been shown to be an ideal method to detect and identify the changes in metabolite profiles in response to an environmental stimuli, in this case nitrogen starvation.

GCT Premier operation in CI mode provides useful data by the generation of pseudo-molecular ions. Exact mass measurements of <5 ppm and the use of i-FIT for matching isotope patterns means that the elemental compositions of the intact derivatized compounds can be readily derived. Therefore the use of CI in conjunction with EI library matching can be a powerful tool in the identification of plant metabolites of unknown structure.

The results presented suggest that nitrate nutrition has wide-ranging effects on plant metabolism, with

nitrogen starvation resulting in a decrease in many amino acids with a concurrent increase in the levels of several carbohydrates.

References

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