

Nota applicativa

## Acidic Herbicides in Drinking Water by LC-MS

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Waters Corporation



This is an Application Brief and does not contain a detailed Experimental section.

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### Abstract

This application brief demonstrates analysis of acidic herbicides in drinking water by LC-MS.

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## Introduction

Compounds analysed in this study are:

1. picloram
2. chloramben
3. 4-nitrophenol bentazon
4. 2,4-D
5. MCPA
6. 2,4,5-T
7. dichlorprop
8. MCPP
9. dichlorobenzoic
10. acifluorfen
11. 2,4,5-TP
12. 2,4-DB
13. dinoseb
14. pentachloropheno

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## Experimental

### HPLC Method

Column: XTerra MS C<sub>18</sub> 2.1 x 100 mm, 3.5 μm

Part number: 186000404

Mobile phase A: 15 mM NH<sub>4</sub>COOH, pH 3.4

Mobile phase B: ACN

Injection volume: 20 μL

Flow rate: 0.2 mL/min to MS

Temperature: 35 °C

Detection: MS ESI

Instrument: Alliance 2695, Micromass ZQ

## Gradient

Time (min)	Profile	
	%A	%B
0.0	75	25
9.0	40	60
14.0	40	60
30.0	10	90

Compound	RF (slope)	r <sup>2</sup>	LOQ	RSD*
1. picloram	3.7	0.999	300	16%
2. chloramben	1.8	0.989	200	9.3%
3. 4-nitrophenol				
4. (non-linear above 500)	474	0.990	<100	5.6%
5. bentazon				
6. (non-linear above 300)	181	—	<100	6.1%
7. 2,4-D	51	0.999	100	7.2%
8. MCPA	43	0.980	200	9.2%
9. 2,4,5-T	105	0.999	100	6.3%
10. dichlorprop	105	0.999	100	5.0%
11. MCPP	136	0.992	100	7.0%
12. dichlorobenzoic	64	0.988	100	5.6%
13. acifluorfen				
14. (non-linear above 300)	50	—	100	11%
15. 2,4,5-TP	91	0.997	100	6.6%
16. 2,4-DB	77	0.999	100	6.3%
17. dinoseb				
18. (non-linear above 200)	>500	—	<100	10.7%
19. pentachlorophenol	67	0.998	100	11.7%

\*average of RSD from 4 levels

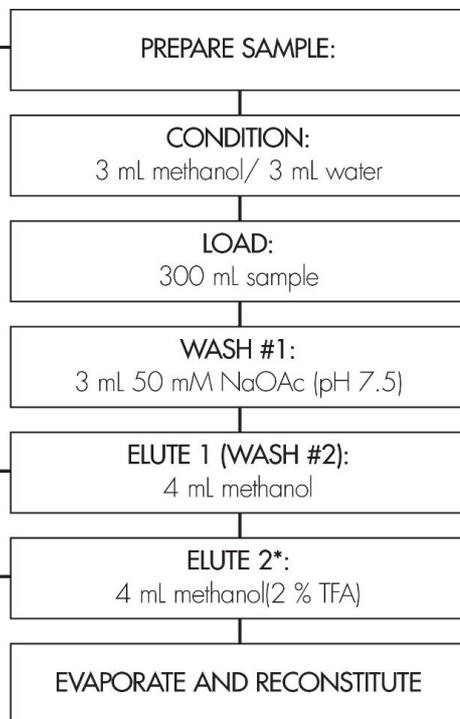
## OASIS® MAX SPE METHOD FOR ACIDIC HERBICIDES

Conditions for Oasis® MAX Cartridge, 6 cc, 150 mg  
Part Number 186000369

Sample is first hydrolyzed at pH 12 for 60 min. Then, pH is adjusted to approx. neutral with HCl before SPE.

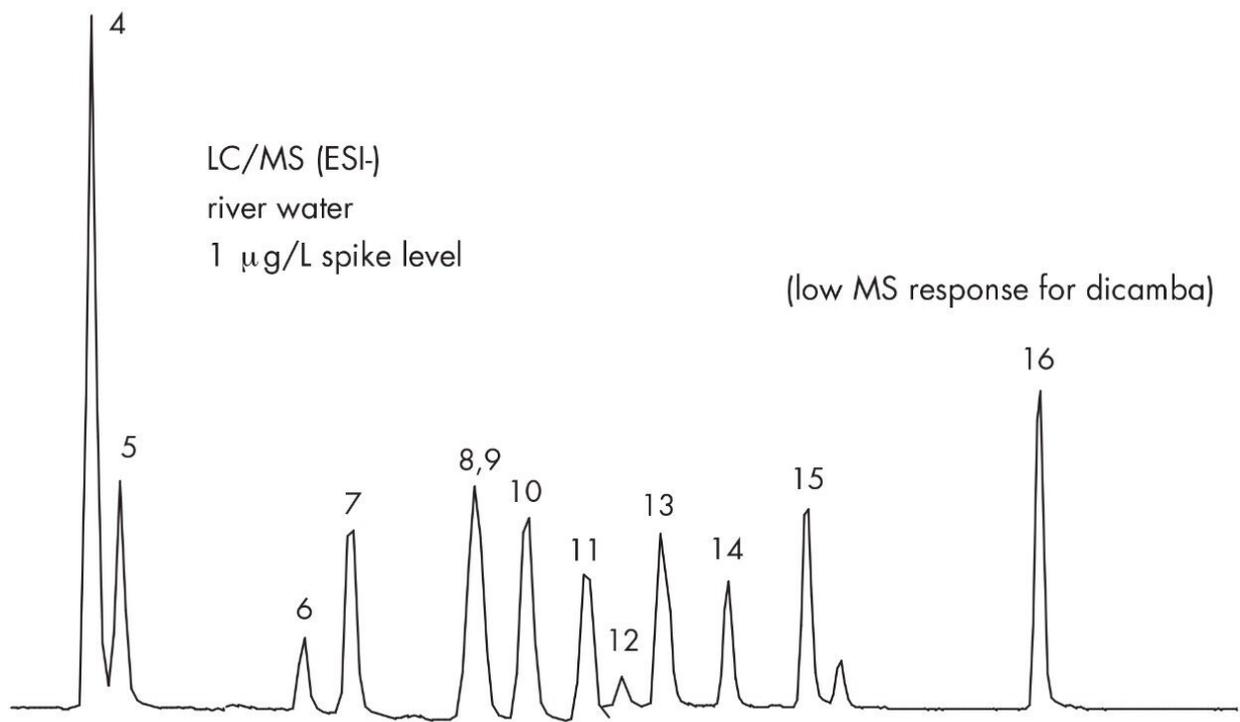
WASH #2 (methanol) will contain bases and neutrals retained by reversed-phase interaction. This fraction may be analyzed for those compounds if desired.

Analytes with  $pK_a < 3$  require strong acid (i.e. trifluoroacetic) at this step.



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## Results and Discussion



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